

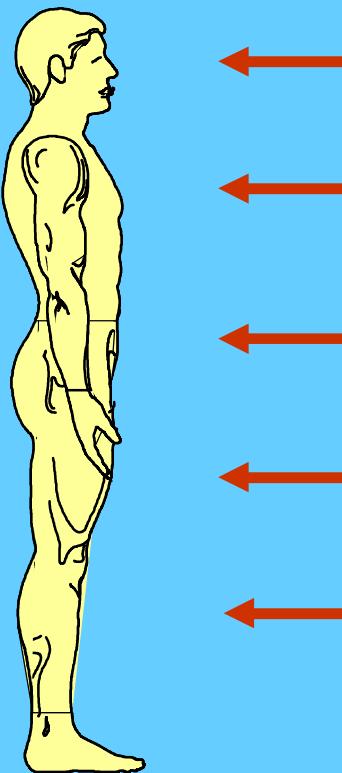
European Work Hazrds Network
29.sept. 2006

Personal Protective Equipment (PPE)

Halvor Erikstein
Occupational Hygienist
Norwegian Union of Energy Workers
SAFE
www.safe.no
halvor@safe.no



Routes of Exposure



Eyes

Respiratory system

Digestion

Skin uptake

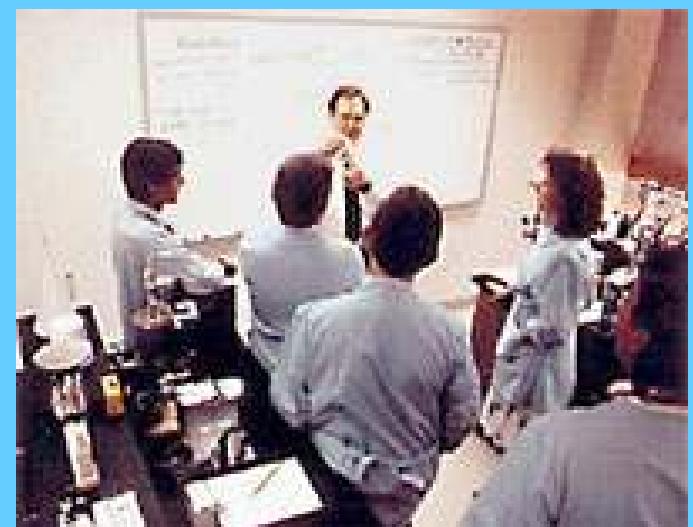
Other routs

Working environment kills



Foto hentet fra boken "The quiet sickness. A photographic chronicle of hazardous work in America"
Earl Dotter, 1998, ISBN 0-932627-85-4. American Industrial Hygiene Association

Chemicals are developed and produced under safe conditions, but where do they end?



Maybe here?



Or here?

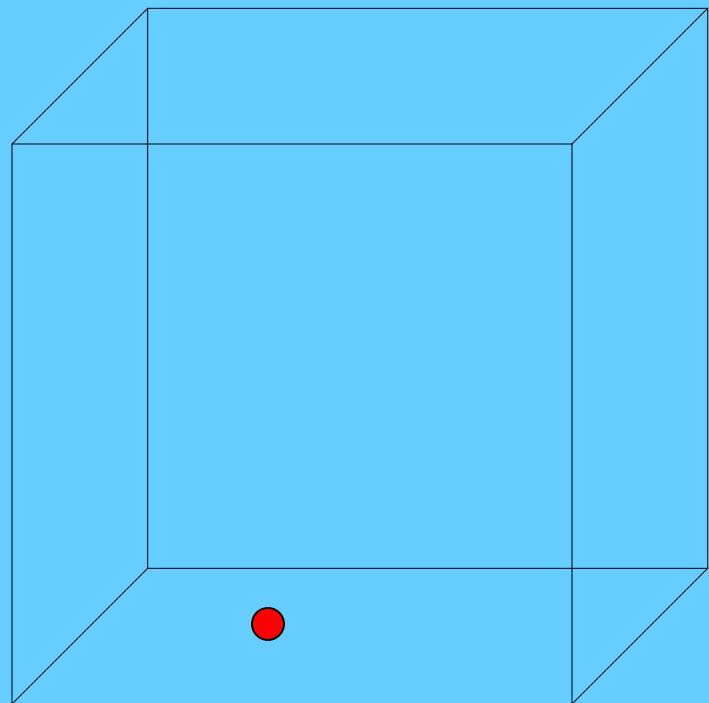


Foto: Halvor Erikstein

Occupational exposure limits (OEL)

parts pr. million (ppm)

$1 \text{ m}^3 = 1000 \text{ litre}$

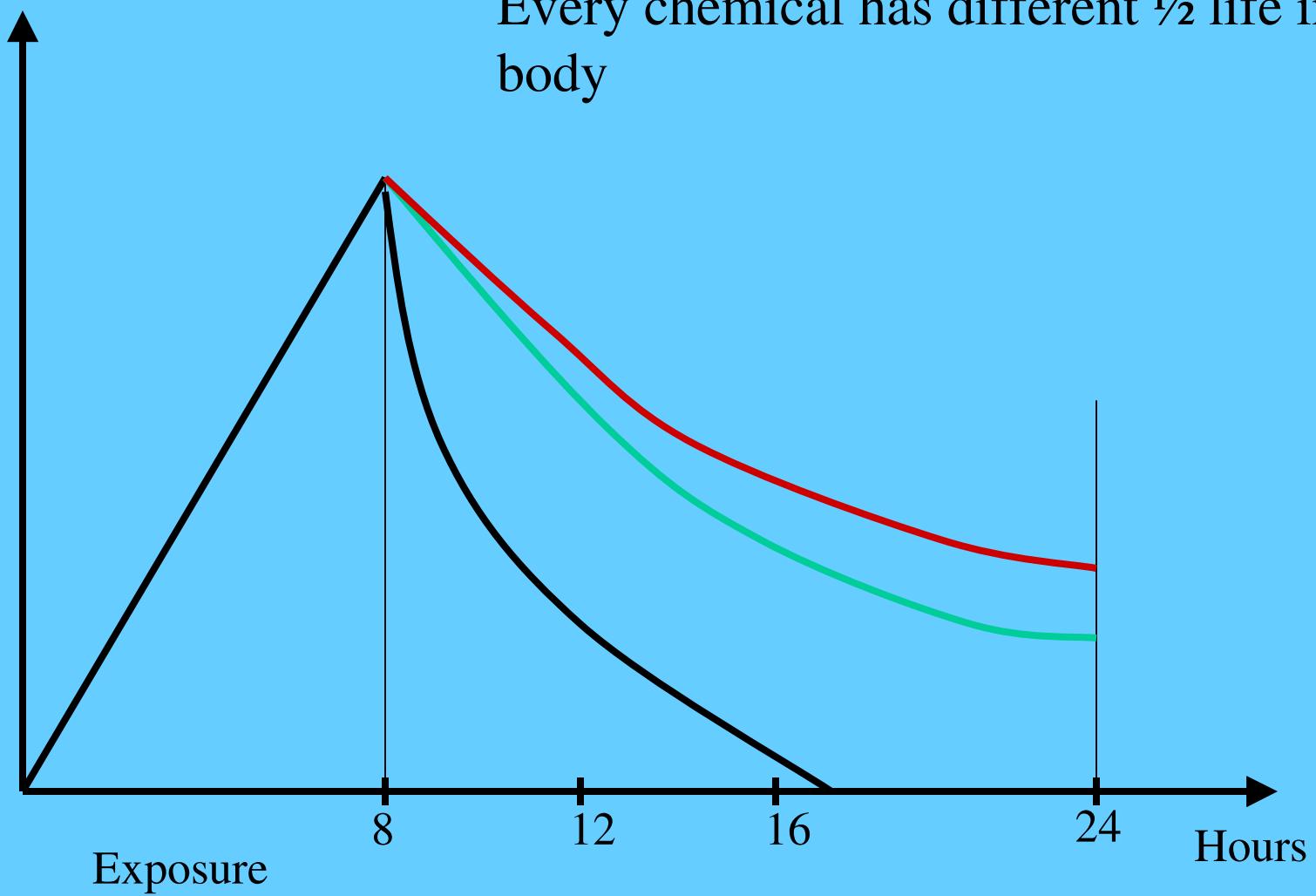


• $1 \text{ volume \%} = 10.000 \text{ ppm}$

• 1 ppm is 1 cm^3 (1 millilitre) diluted in 1m^3 . The weight is in (mg/m^3)

Concentration i body

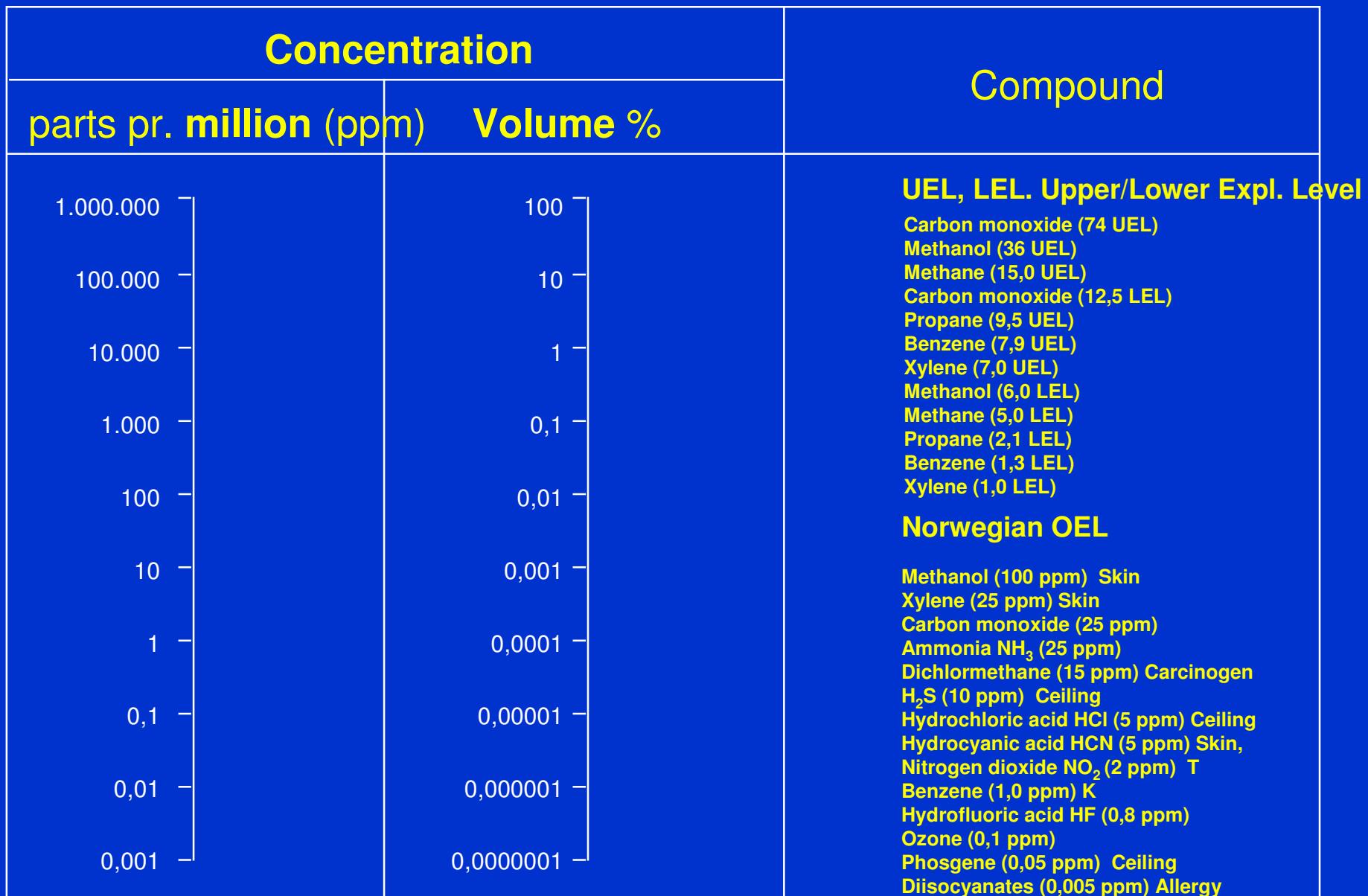
Every chemical has different $\frac{1}{2}$ life in the body



Different Occupational Exposure Level

- TLV (US)
- MAK (Germany)
- OEL (UK)
- Administrative normer (Norway)

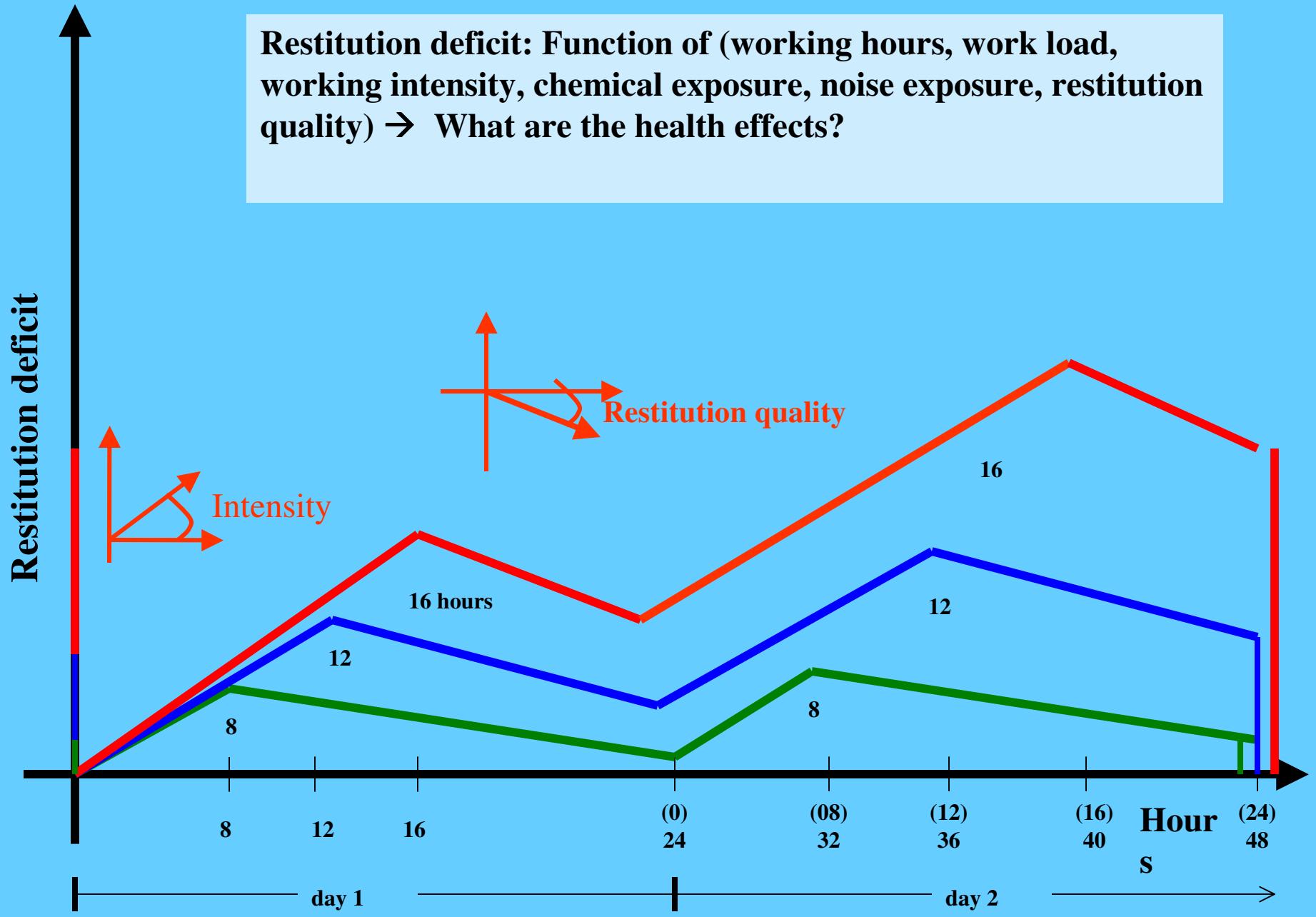
The Hazard ladder

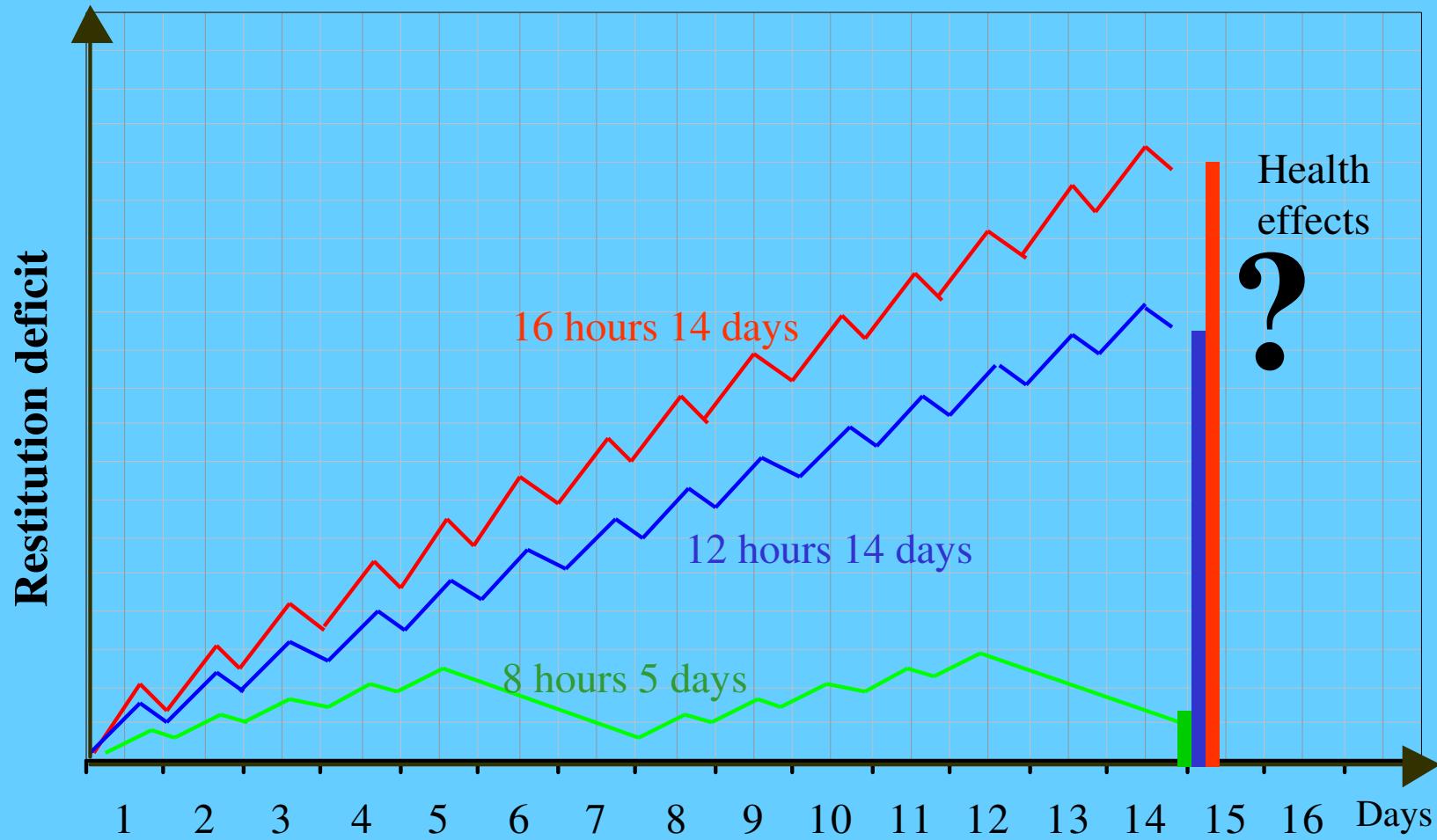


Restitution deficit will be a function of;

- Work length 8 / 12 / 16 / hours
- Number of subsequent days of working
- Day/night, circadian rhythme
- Work load
 - Work intensity
 - Work load
 - Psychological factors
 - + + + + ? ?
- Restitution quality
 - Duration of restitution 8 / 12 / 16/ hours...
 - Sleep quality, sleep length, circadian rhythme
 - + + + ? ?
- Exposure
 - Chemicals
 - Noise
 - + + + ? ?

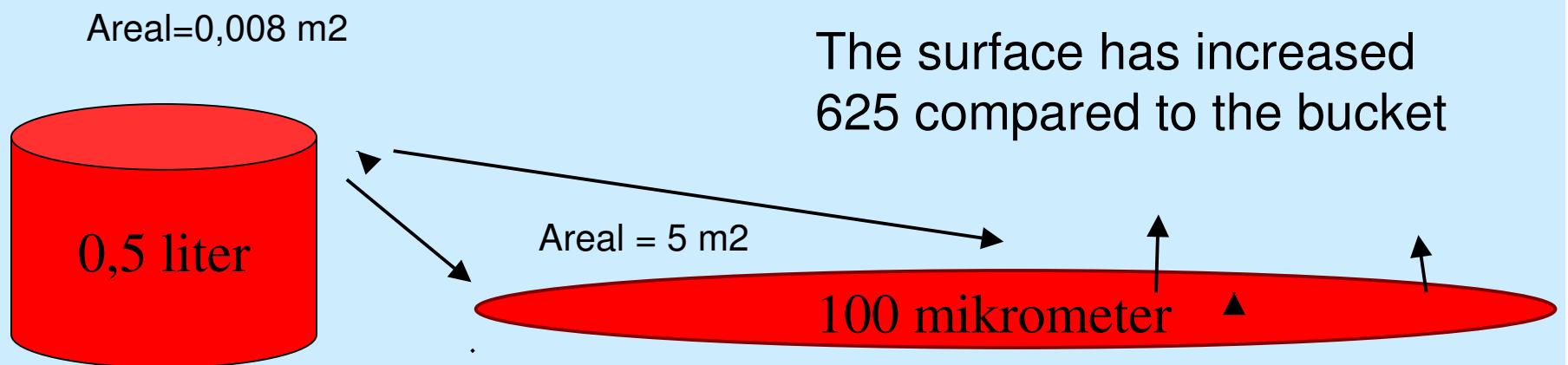
And a combination of everything!!





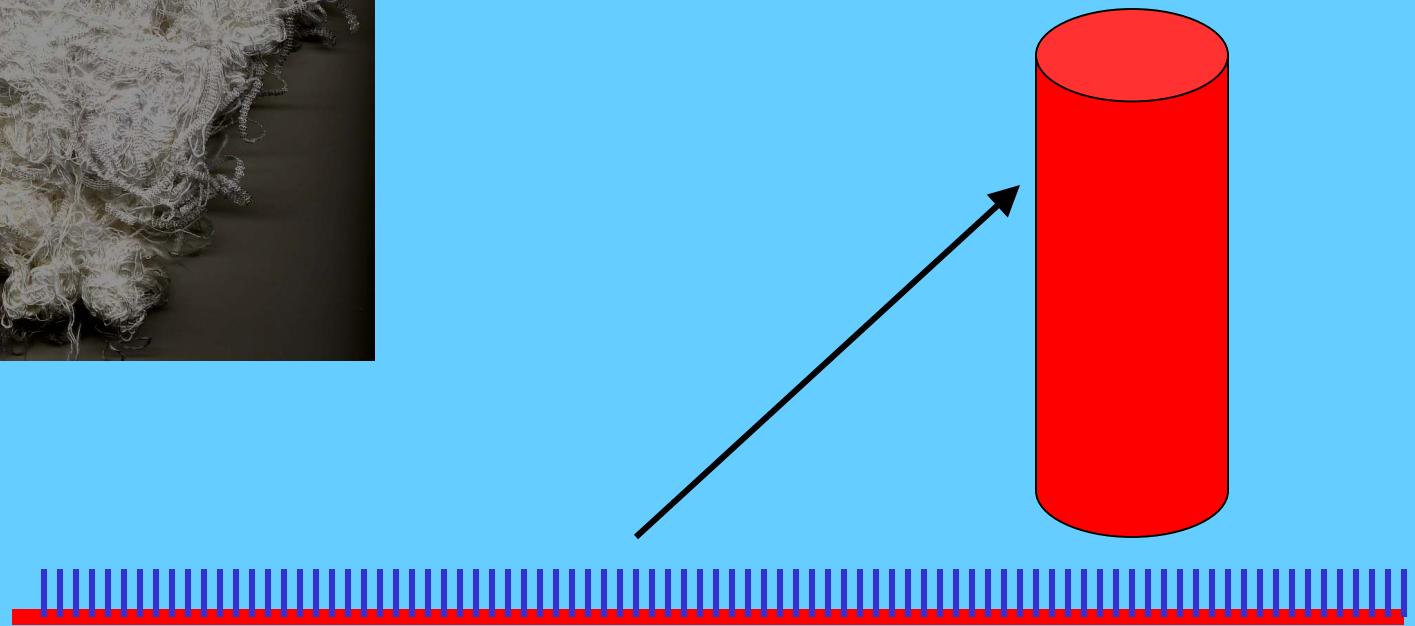
Source strength

Increase in surface

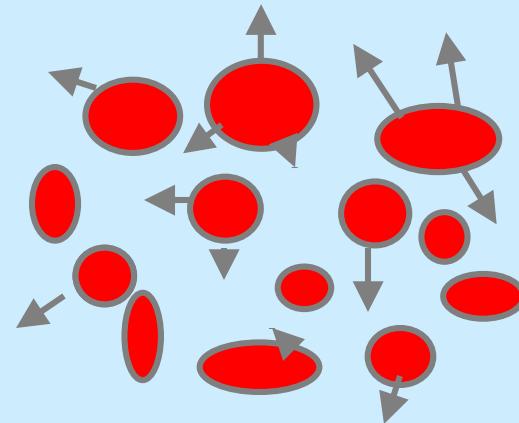
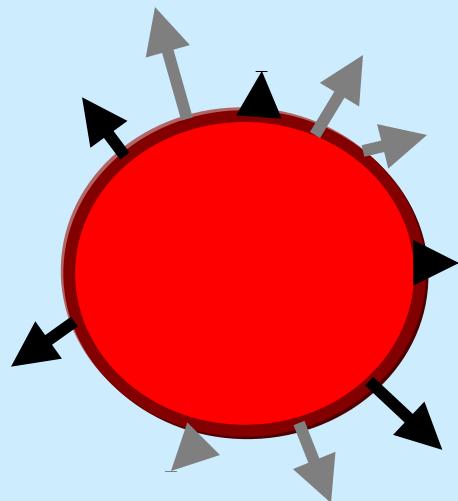




Extreme increase in
surface when the
chemical is spread on
the fibers



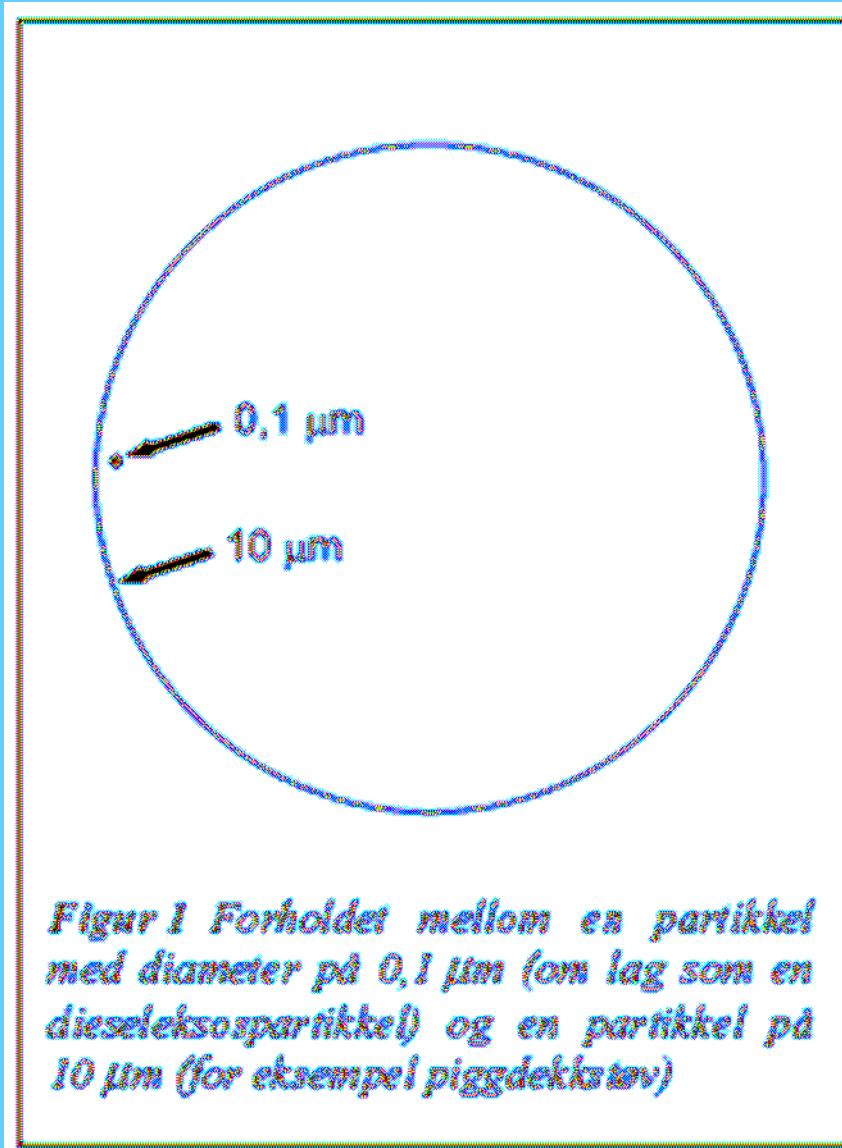
Aerosols



- 1cm³ split to 2 mikrometer increase the surface 10.000.000 times

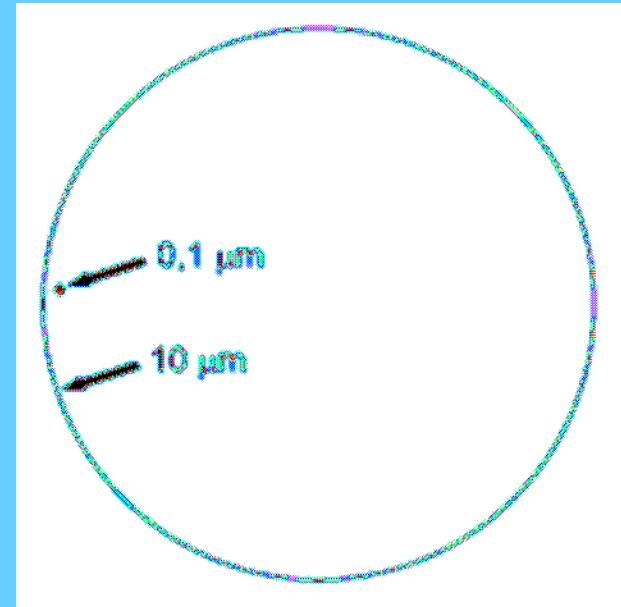
Seize of aerosols

- A person can spot a single particle at about 50 micrometer.
- Smaller particles are only visible in strong light.
- Particles less than 10 micrometer will be seen as fog.

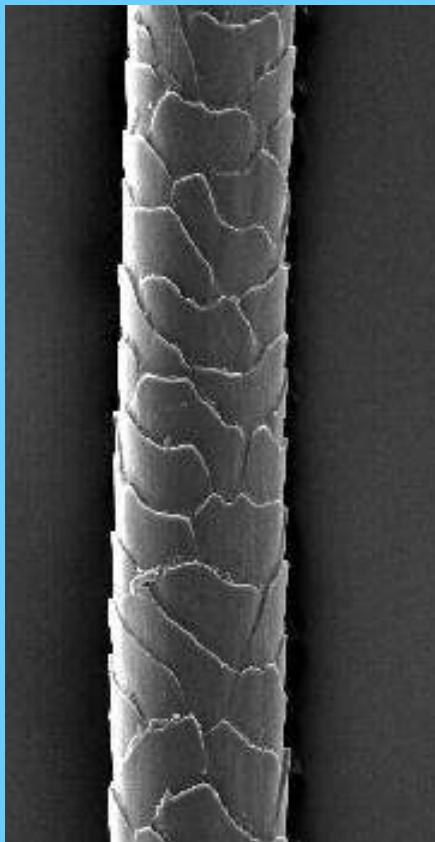


Relative surface and relative number
for a given weight of spherical
particles with different diameter

Diameter Micrometer	Relative surface	Relative number
10	1	1
2,5	4	64
1,0	10	1000
0,1	100	1000.000



Seize of aerosols II



- Hair **60 –100 mikrometer.**
- Bacteria **0,3 – 50 mikrometer**
- Virus **0,01 - 0,05 mikrometer.**
- Particles with aerodynamic diameter less than **10 mikrometer** are suspended dust

Aerosol science for industrial hygienists

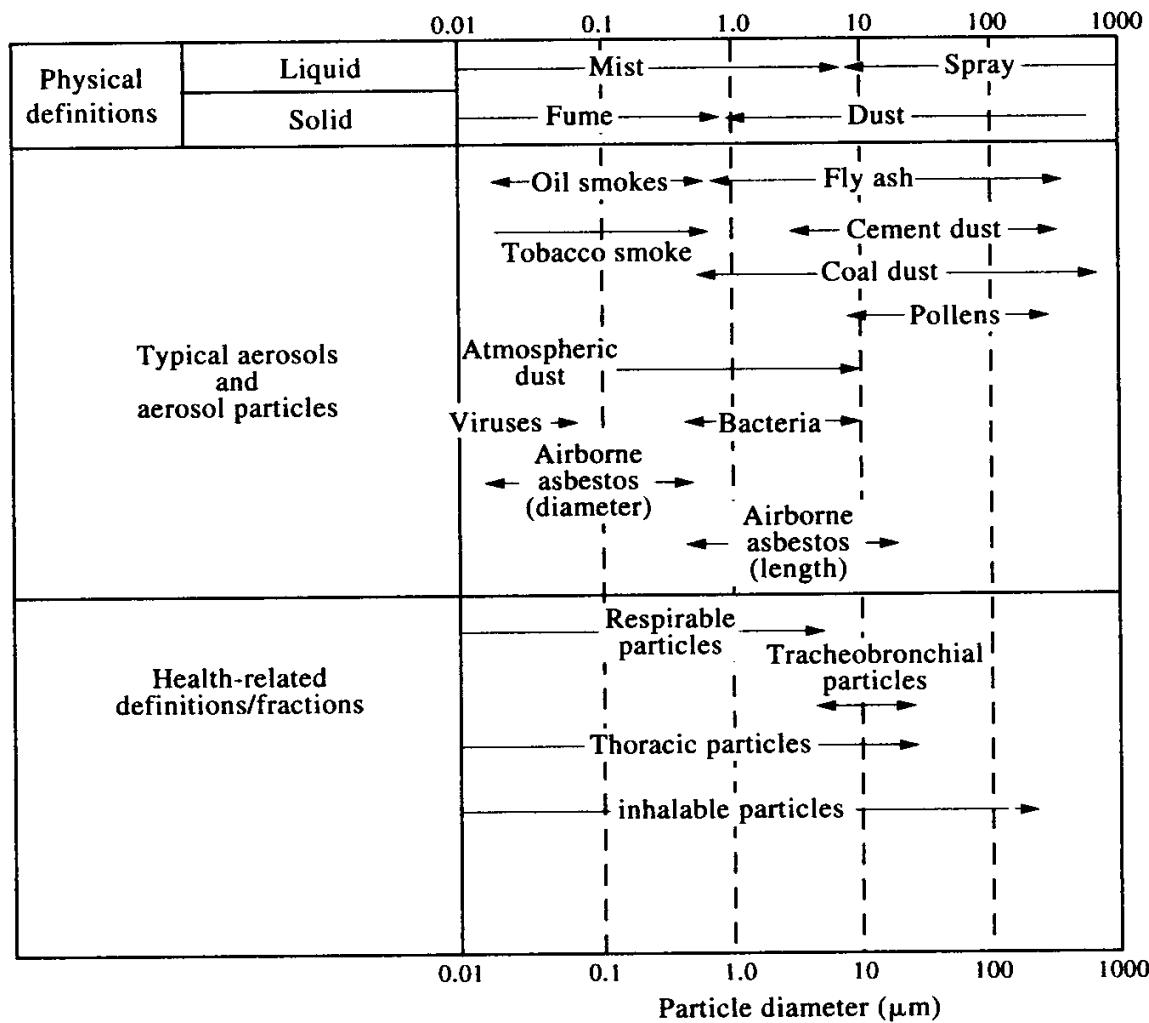


Figure 1.1. Summary classification of aerosols (from Vincent, J.H., *Aerosol Sampling: Science and Practice*, Copyright 1989, adapted by permission of John Wiley and Sons Limited).

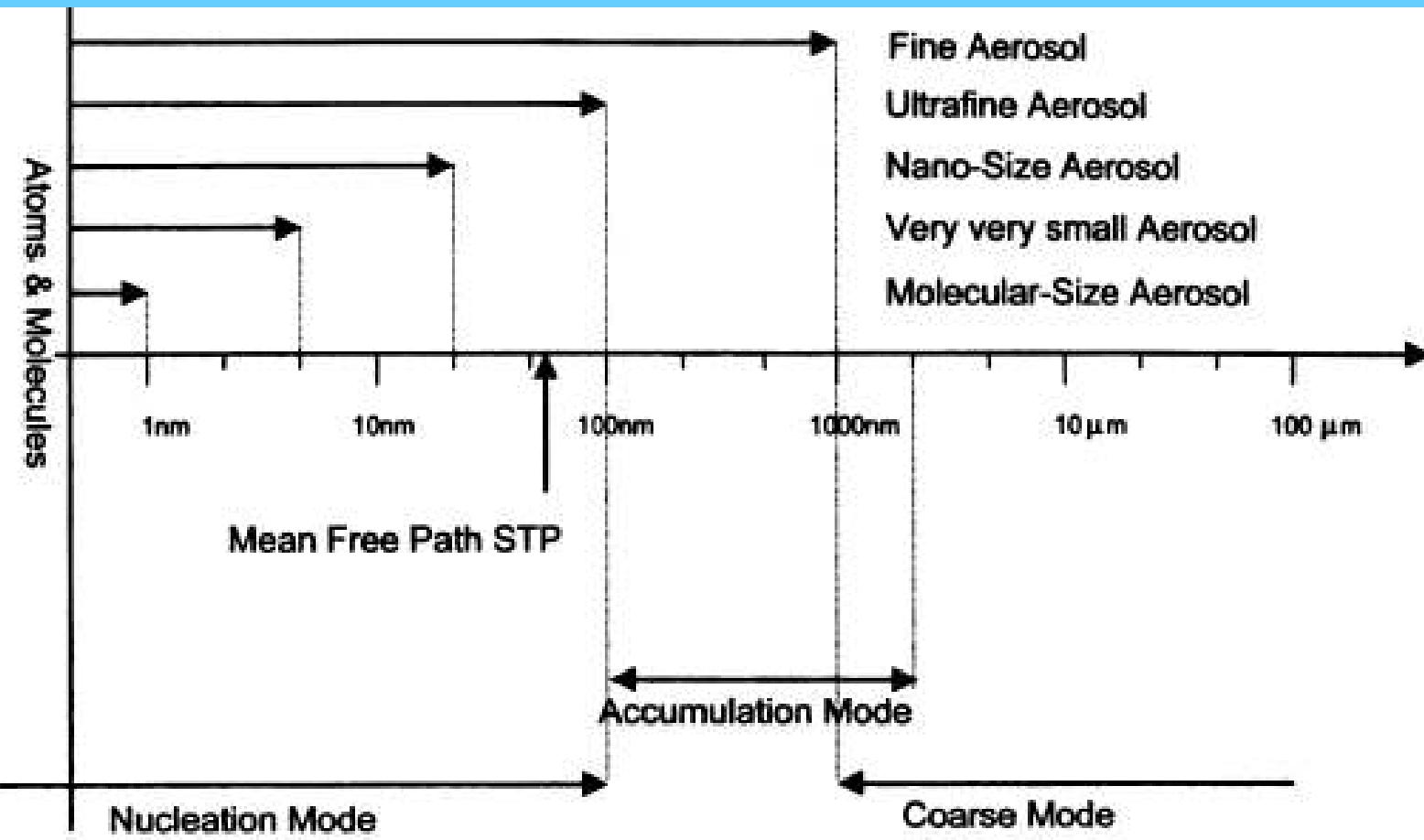


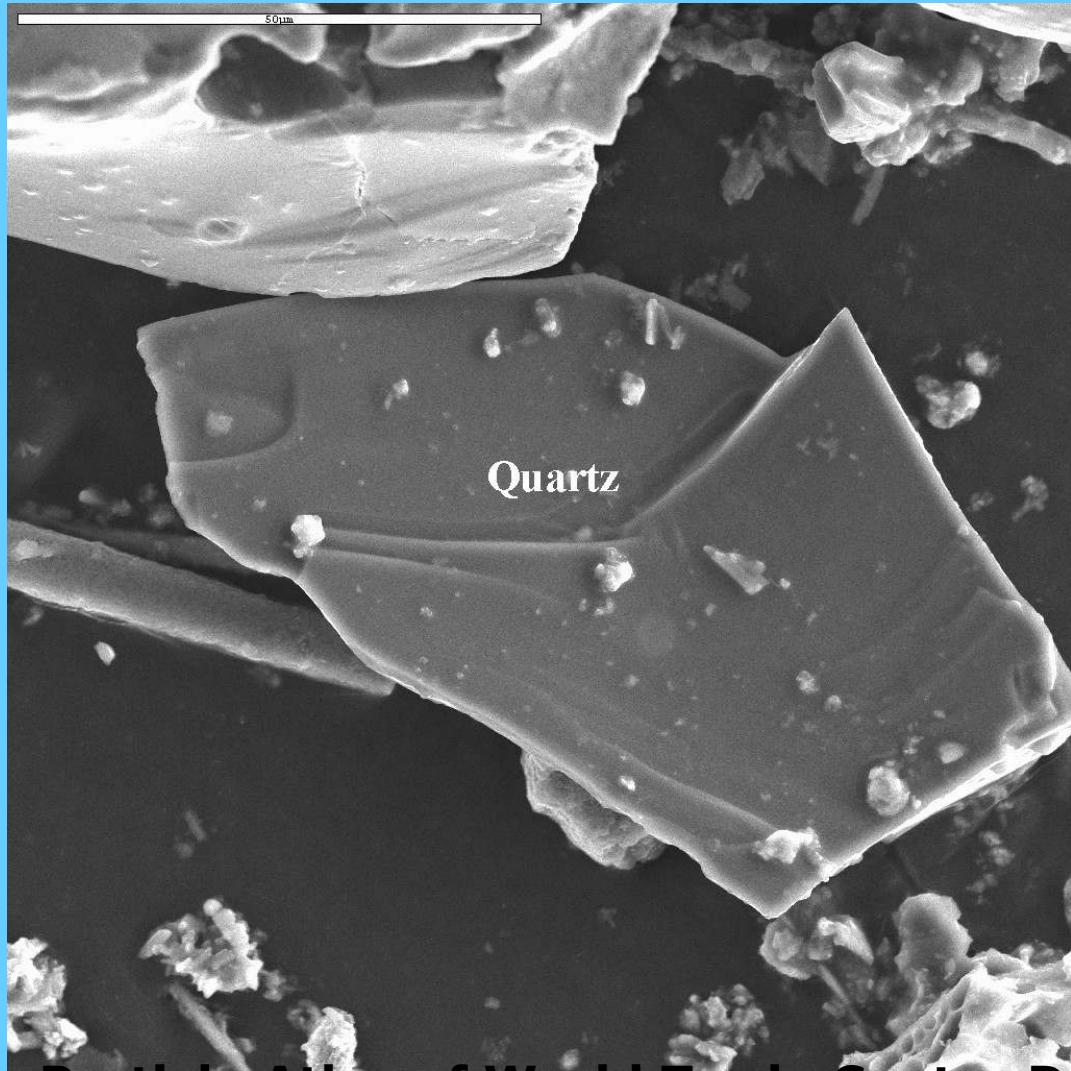
Figure 3.

The particle size classes: **coarse mode**, particles larger than about 1 μm mainly produced by diminution processes; **fine aerosol**, particles smaller than about 1 μm mainly built up by nucleation, condensation and coagulation; **nucleation mode** and **ultrafine aerosol**, particles smaller than about 100 nm; **nanosized aerosol**, particles smaller than about 20 nm; **very very small aerosol**, particles smaller than about 5 nm, particle behaviour dominated by surface effects, total number of molecules less than 500, **molecular size aerosol**, particles smaller than about 1 nm, less than 10 molecules in the particle. Reproduced from Preining (1998).

How fast will an aerosol fall in the air?

• Aerodynamic diameter (mikrometer)	Velocity metre/hour
100	1080
40	172
10	11
5	3
1	0,11

Silica dust



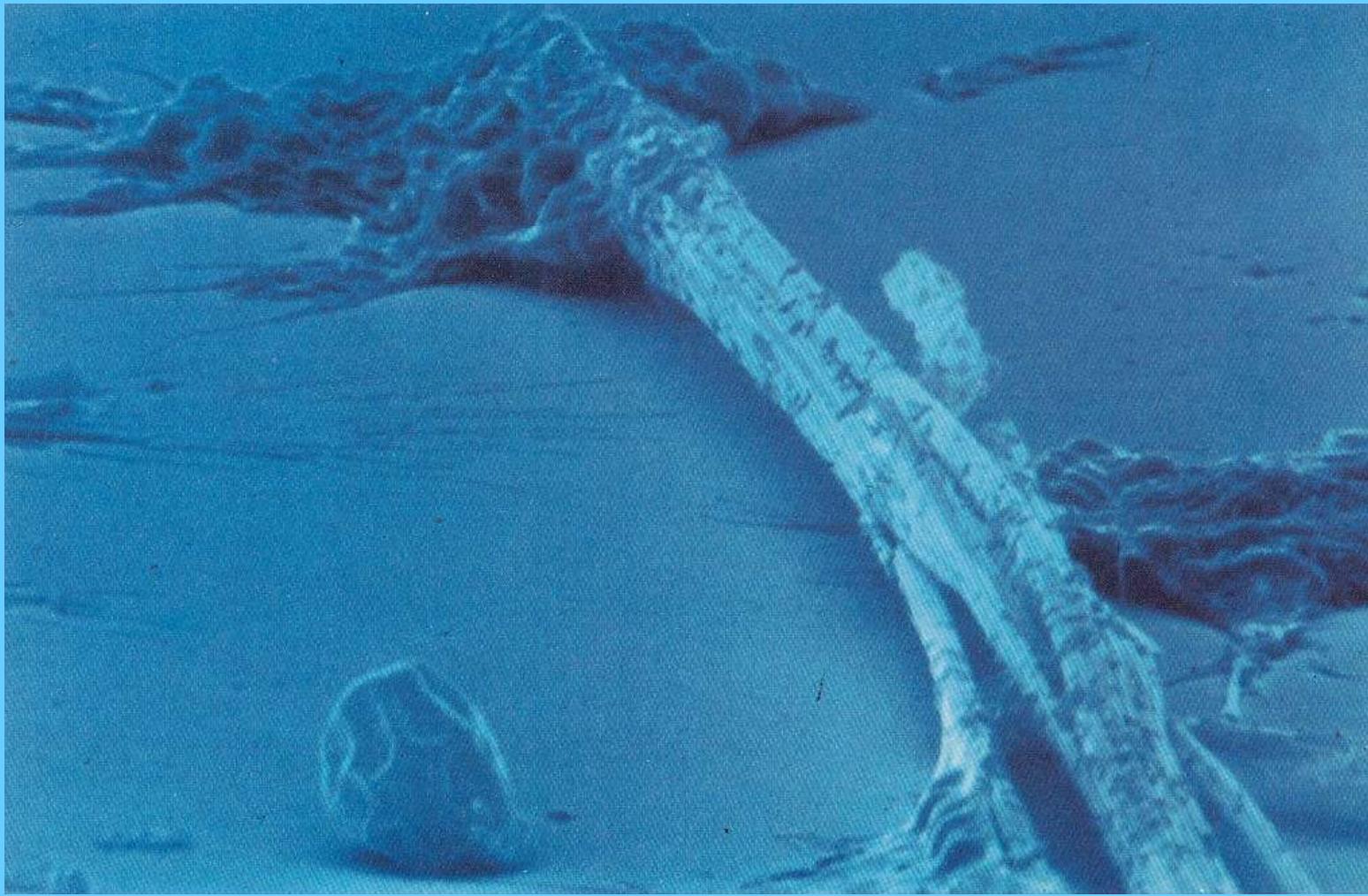
Particle Atlas of World Trade Center Dust

http://pubs.usgs.gov/of/2005/1165/table_1.html

Very
dangerous,
but
forgotten?...



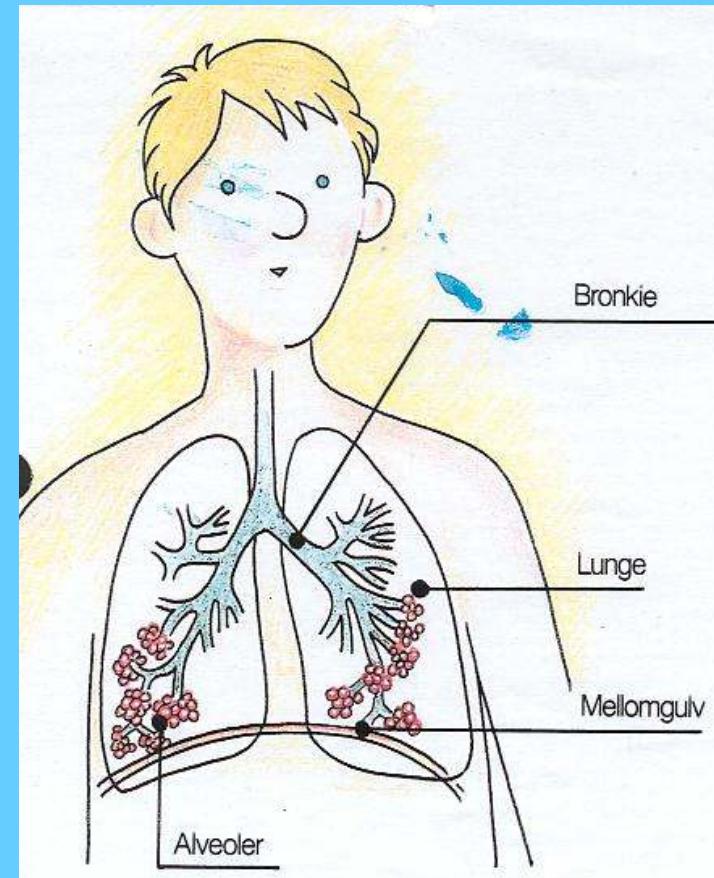
Asbestos – still a major problem



The seize of the particles decies where they ends

**Seize
(mikrometer)**

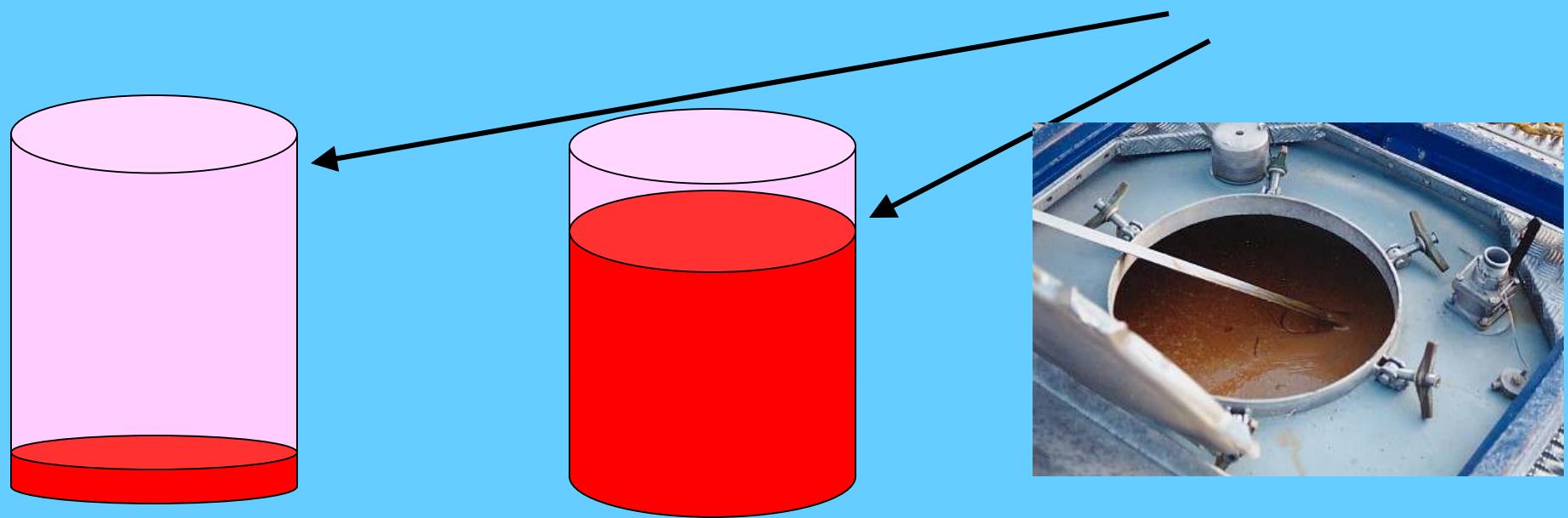
10 - 5	Nose and throat
5 - 3	Windpipe
3 - 2	Bronchus
2 - 1	Bronchus
1 - 0,1	Alveols
< 0,1	Kan pustes ut igjen



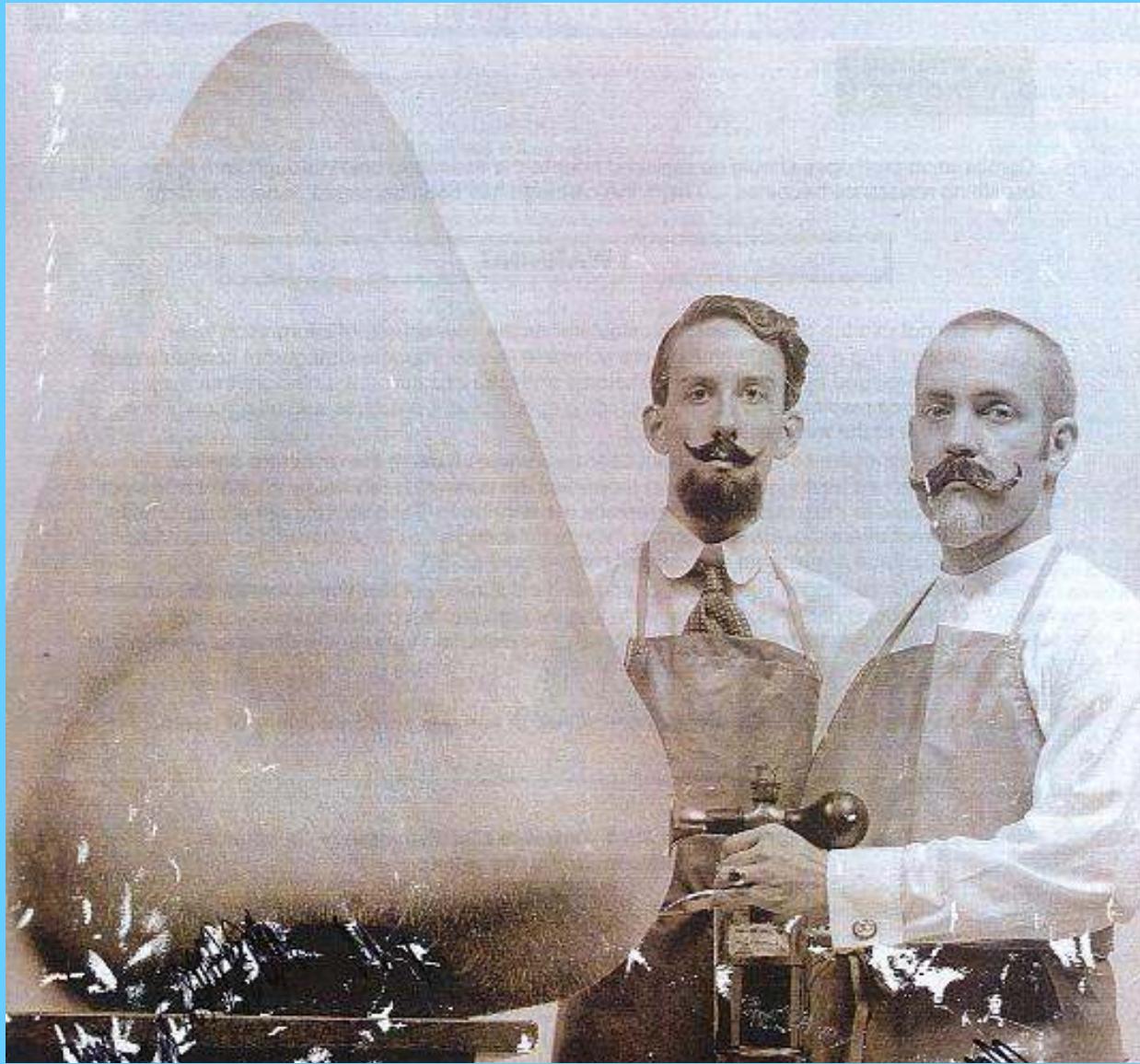
The rule of 1300

Toluene has a vapor pressure of 20 mmHg.

Eksempel: $20 \text{ mmHg} \times 1300 = 26000 \text{ ppm}$



Odor Thresholds

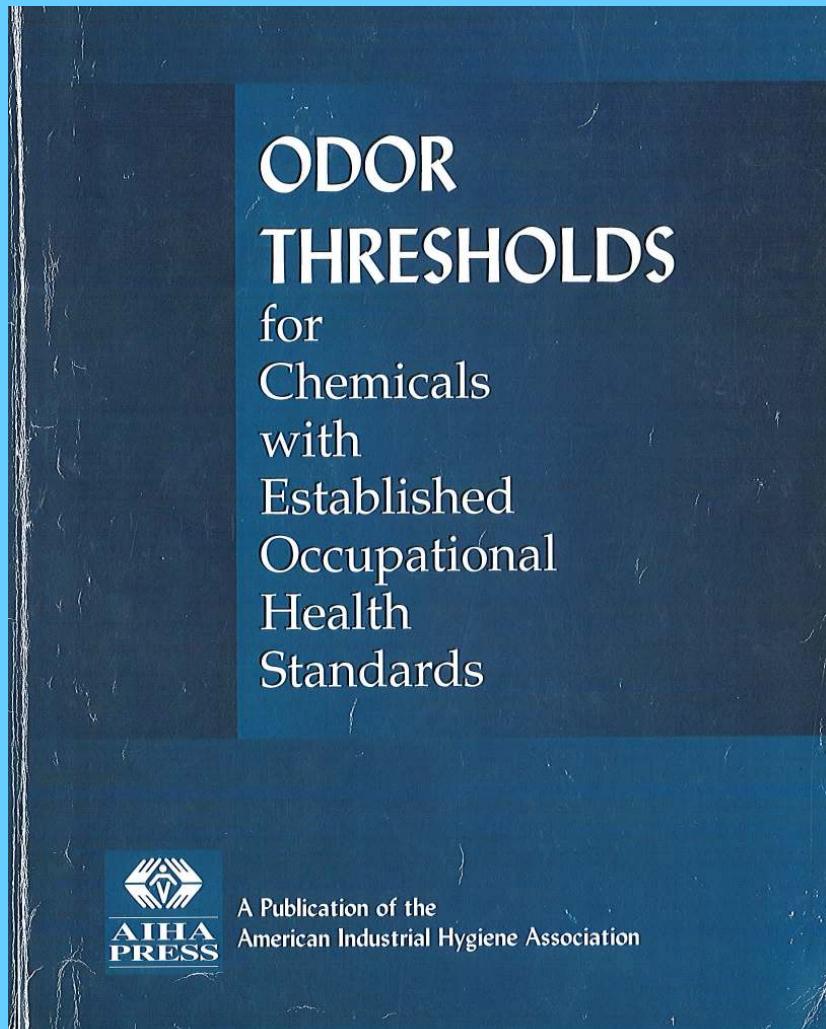


Odor Thresholds

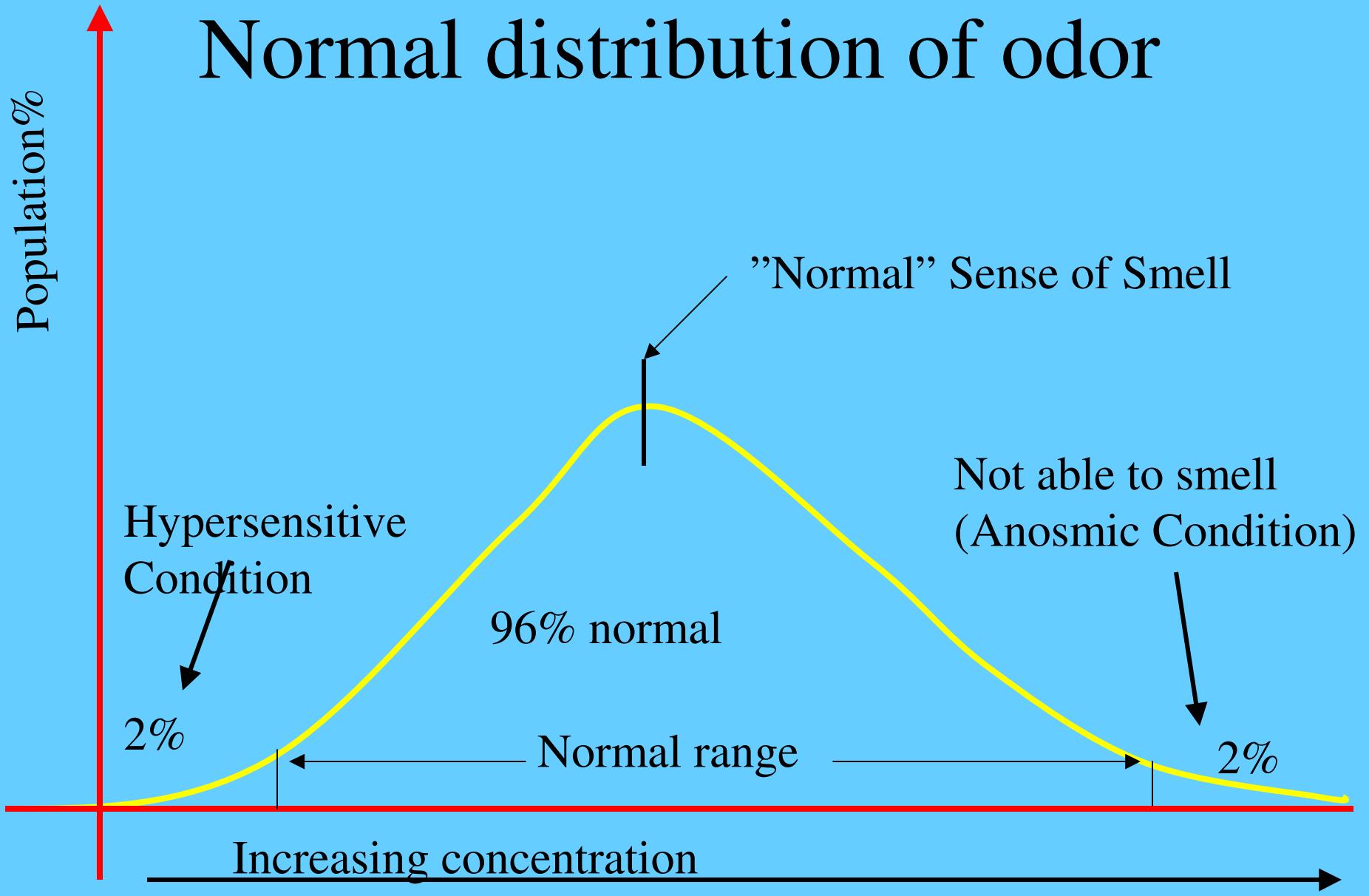
**Odor Thresholds for
Chemicals with
Established
Occupational Health
Standards.**

**American Industrial
Hygiene Association,
1995.**

ISBN 0-932627-34-X



Normal distribution of odor



Some odor thresholds

Chemical OEL* (ppm)	Low	High	Geometric mean
*Norwegian			
Dichormethane (15)	1,2	440	160 d
Styrene (25)	0,017	1,9	0,14 d
Methanol (100)	4,2	5960	160 (all ref.)
Xylene (25)	0,06	40	20d
Hydrochloric acid (5)	0,256	10,1	Not accepted
Formaldehyde (0,5)	0,027	9770	Not accepted
Isopropyl Alcohol	37	610	43
Ammonia NH ₃ (25)	0,04	53	17 d
Acetone (125)	3,6	653	62 d
Toluene diisocyanate TDI (0,005ppm)			0,2-0,4*
Benzene (1)			2,14-12 **
Toluene (25)	0,16	37	1,6

Odor Thresholds for Chemicals with Established Occupational Health Standards. American Industrial Hygiene Association, 1995. ISBN 0-932627-34-X

*<http://www.bASF.com/businesses/polymers/urethanes/pdfs/chemicals/Other/2000tdihandbook.pdf>

** Maslansky and Maslansky, Health and Safety at hazardous waste Sites, 1997, ISBN 0-442-02398-7, side 102

- Two percent of the population are predictably hypersensitive, and two percent insensitive.
- The insensitive range include people who are **anosmic** (unable to smell) and **hyposmic** (partially unable to smell).

The difference between people

- The sensitive range includes people who are hyperosmic (very sensitive) and people who are sensitized to a particular odor through repeated exposure.
- Individual threshold scored can be distributed around the mean value to several orders of magnitude.

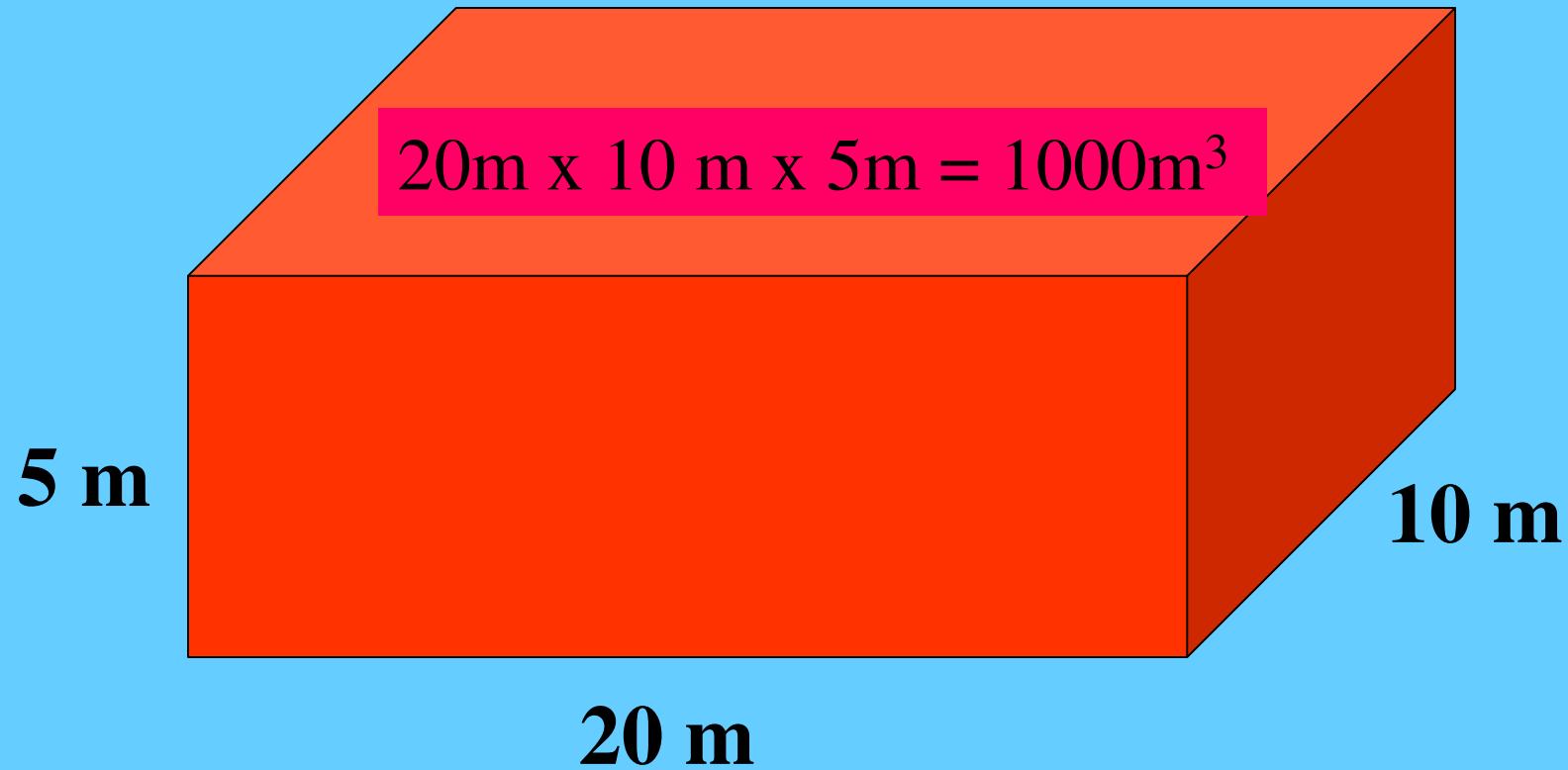
Variation

- A person may be hyposmic to one odorant and hyperosmic to another.
- The variation occurs in specific anosmia and is often caused by repeated exposure to a particular odor. It is not uncommon among chemists or other workers who have had daily exposure to an odorant over a period of years

Odor fatigue

After 3 min of exposure to an odorant, the subject's perceived intensity of the odorant is reduced about 75%

Concentration



Room volume 1000 m^3

1 kilogram TOLUENE

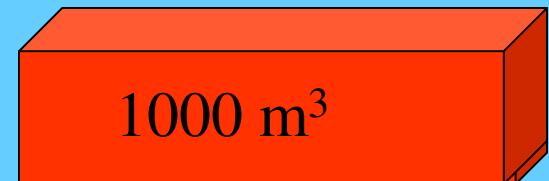
$1 \text{ kg} = 1000 \text{ gram} = 1000000 \text{ milligram}$

$1 \text{ ppm} = 3,83 \text{ mg/m}^3$

(1000 mg/m^3) .

Concentration in the room

$1000 \text{ mg/m}^3 / 3,83 \text{ mg/m}^3 = 261 \text{ ppm}$



Air volume necessary to dilute to occupational exposure level

- Norwegian OEL (25 ppm, 94 mg/m³)?
- **10000000 mg / 94 mg/m³ = 10640 m³**



Lower Explosive Level working in confined space

- 1,1 volume % = 11000 ppm
- 11000 ppm x 3,83 mg/m³ = 42130 mg/m³
- 1000000 mg / 42130 mg/m³ = 23,7 m³



LEL from: NIOSH Pocket Guide to Chemical Hazards
<http://www.cdc.gov/Niosh/npg/npgd0619.html>

MSA Cartridge Life Calculator



www.msanet.com/international/

Search | 0 blocked | ABC Check | AutoLink | AutoFill | Options

Select Region

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Regulatory Info
Tools >>>
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Glossary

MSA International

- Response Respirator Selector
- Cartridge Life Calculator
- Respirator Change Test Data
- Product Image Library

With international affiliates, serving more than 120 countries around the globe, A dedication to global product development allows MSA to better serve customer needs the same way all around the world. Our Affiliates obtain and market products from MSA manufacturing and research centers on all continents, as well as their own countries.

All major MSA manufacturing locations have received Quality System certification to ISO 9000 or equivalent. MSA products generally carry Product Approvals and Certifications, ranging from NIOSH, MSHA, CE, CSA, MET, UL, etc., as well as local approvals. In addition, MSA companies offer local or regional repair and maintenance service.

MSA International Headquarters in Pittsburgh, Pennsylvania, USA, offers exports of U.S. produced products both directly and through numerous distribution partners. MSA International also manages all MSA Affiliate operations outside of North America and Europe. The people of MSA International stand ready to solve your safety equipment needs!

Mission. So that men and women may work in safety and that they, their families and their communities may

Three small square images on the right side of the page. The top one shows firefighters in a smoky environment. The middle one shows a soldier in a desert setting. The bottom one shows a worker in a industrial setting.

<http://www.msanet.com>

Gjennombruddstid til 10 ppm ved 261 ppm toluen

MSA Cartridge Life Expectancy Results

Final Breakthrough Time Calculation

When using a ComfoClassic Facepiece with a Comfo GMA Cartridge under the following conditions:

Chemical Name:	Toluene
Chemical PEL (ppm):	200 OSHA PEL
Temperature:	20°C
Relative Humidity:	40 %
Pressure:	760 Torr
Breathing Rate:	60 LPM
Use Concentration:	261 ppm
Breakthrough Concentration:	10 % OSHA PEL

**The estimated
Breakthrough Time
at which cartridges
need to be replaced is:**

377 minutes

"NaN" = Not A Number

For example, a result cannot be calculated if Use Concentration is less than Breakthrough Concentration.

**40 %RH
Breakthrough Time 377 min**

Back to the Calculator

MSA Cartridge Life Expectancy Results

Final Breakthrough Time Calculation

When using a ComfoClassic Facepiece with a Comfo GMA Cartridge under the following conditions:

Chemical Name: Toluene
Chemical PEL (ppm): 200 OSHA PEL
Temperature: 20°C
Relative Humidity: 60 %
Pressure: 760 Torr
Breathing Rate: 60 LPM
Use Concentration: 261 ppm
Breakthrough Concentration: 10 % OSHA PEL

**The estimated
Breakthrough Time
at which cartridges
need to be replaced is:**
303 minutes

"NaN" = Not A Number

For example, a result cannot be calculated if Use Concentration is less than Breakthrough Concentration.

60 %RH
Breakthrough Time 303 min

[Back to the Calculator](#)

MSA Cartridge Life Expectancy Results

Final Breakthrough Time Calculation

When using a ComfoClassic Facepiece with a Comfo GMA Cartridge under the following conditions:

Chemical Name: Toluene
Chemical PEL (ppm): 200 OSHA PEL
Temperature: 20°C
Relative Humidity: 80 %
Pressure: 760 Torr
Breathing Rate: 60 LPM
Use Concentration: 261 ppm
Breakthrough Concentration: 10 % OSHA PEL

**The estimated
Breakthrough Time
at which cartridges
need to be replaced is:**
200 minutes

"NaN" = Not A Number

For example, a result cannot be calculated if Use Concentration is less than Breakthrough Concentration.

**80 %RH
Breakthrough Time 200 min**

[Back to the Calculator](#)

JANUARY 16, 1999

6-inch growth could qualify him for the Guinness World Record

Man grows world's longest nose hair!

BUFFALO, N.Y. — As gross as it might seem, Alan Byrdal is about to make it into *Guinness World Records* — for having the longest nose hair in the world!

The 64-year-old former pool table salesman has been pursuing the record "since the mid-1980s," he says. And now that the longest of his nose hairs has reached a staggering 6.12 inches, he's ready to stake his claim to the record.

"I guess everyone wants to be known for something, and this is it for me," he said. "I've heard all the comments — I know some people might say it's disgusting or whatever. But you can bet most of those people will never make it into any kind of record book."

Byrdal said he had to start cutting his nose hairs when he was about 27 years old because they grew so fast.

"My dad and grandfather had the same problem," he said. "Mine started growing when I was fairly young and back then it was pretty gross. Long nose hairs are not really the kind of thing the girls go for."

But when Byrdal reached his 50s, he decided to forget

By AMY LECHNER
Weekly World News

about trimming his nose hairs every day.

"By that time my hairs were growing faster than they ever had," the lifelong bachelor said. "I decided, to heck with it, let's see how long they can grow."

Byrdal now admits he "probably became obsessed" with his bizarre hobby.

"It became the most important thing in my life," he said. "I still remember how surprised I was when I found out there wasn't an official world record for longest nose hairs."

Byrdal said he probably would never have done it if he had the active social life he enjoyed when he was younger.

"Nowadays I live by myself and hang around the same group of guys most of the time," he said. "I guess the girl down at the grocery store has seen me so many times she doesn't even notice anymore."

"But I'm getting ready for that all to change now. I'm going to be famous when people all over the world hear about how long my nose hairs are."

SWEET SMELL OF SUCCESS
After more than 15 years, Byrdal has almost reached goal of getting into *Guinness World Records*.

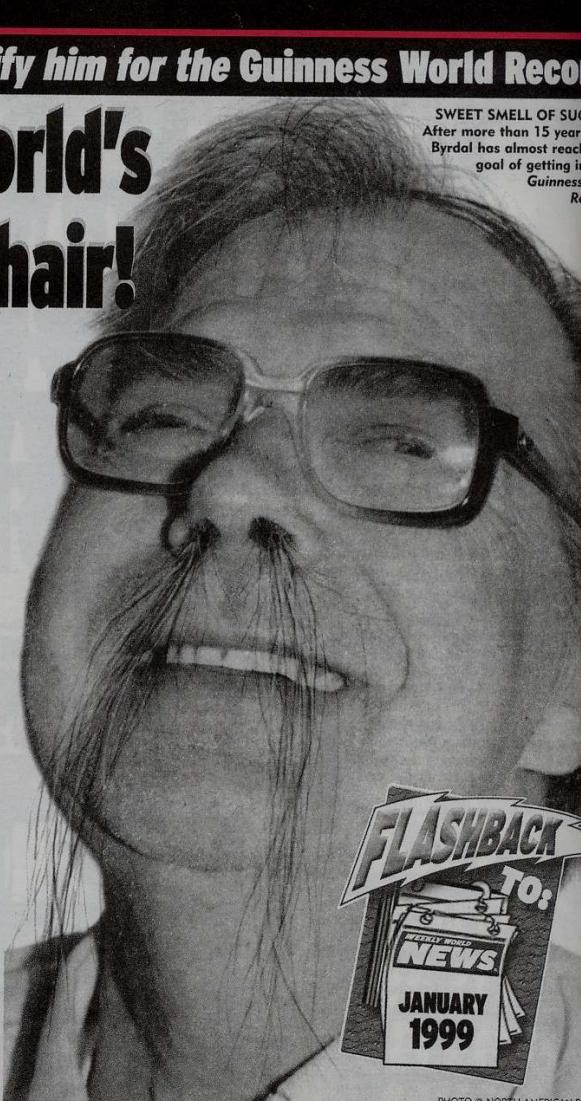


PHOTO © NORTH AMERICAN PRESS

Inspired by Alan Byrdal's courage and passion, Armie Boyd has cultivated the world's longest ear hair in hopes of gaining his turn in the spotlight.

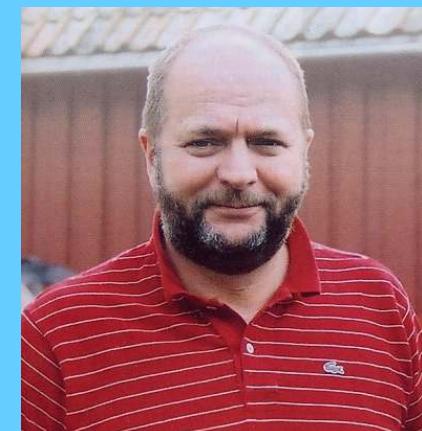
"At first I just had these monster earfros," Boyd says of the experience of developing his hirsute auricles. "Then I made them into some kicking dreadlocks, but that didn't fly so well with the boss—I work in accounting. So finally I bought myself some beads and had my mother braid them."

WWN EXTRA

WORLD'S LONGEST EAR HAIR!



Limitations
for filter
masks:
No facial
hair



Laboratory testing of beard vs. clean shaven



- | | |
|---|--|
| <ul style="list-style-type: none">• Clean shaven• Half mask• Full face maske• Beard• Half mask• Full face mask | <p>x times cleaner inside</p> <ul style="list-style-type: none">• 2950• > 10.000• 12 half mask,• 30 |
|---|--|

The test perform in laboratory under perfect conditions. No relevance to normal use, but shows how facial hair are compromising negative pressure filter masks

Effect of Facial Hair on the Face Seal of Negative-Pressure Respirators.

Am. Ind. Hug. Assoc. J. 45(1):63-66 (1984).

O.T. Skredtvedt and J.G. Loschiavo

Assigned Protection Factor

The assigned protection factor (APF) of a respirator reflects the level of protection that a properly functioning respirator would be expected to provide to a population of properly fitted and trained users. For example, an APF of 10 for a respirator means that a user could expect to inhale no more than one tenth of the airborne contaminant present.

<i>Respirator Class and Type</i>	<i>OSHA Cadmium Std.</i>	<i>NIOSH</i>
Air Purifying		
Filtering Facepiece	10	10
Half-Mask	10	10
Full-Facepiece	50	50
Powered Air Purifying		
Half-Mask	50	50
Full-Facepiece	250	50
Loose Fitting Facepiece	25	25
Hood or Helmet	25	25
Supplied Air		
Half-Mask-Demand	10	10
Half-Mask-Continuous	50	50
Half-Mask-Pressure Demand	1000	1000
Full-Facepiece Demand	50	50
Full-Facepiece Continuous Flow	250	50
Full-Facepiece Pressure Demand	1000	2000
Loose Fitting Facepiece	25	25
Hood or Helmet	25	25
Self Contained Breathing Apparatus (SCBA)		
Demand	50	50
Pressure Demand	>1000	10,000

Assigned Protection Factor

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<http://www.ehso.com/RespProtectionSelection.htm>

Electrostatic filter



Mechanical filter



Electrostatic versus mechanical filters

- A study conducted by the INRS (National Research and Safety Institute) has found that certain types of particulate filter respirators lose their effectiveness over time.
- Filtration levels for solid and liquid aerosols are currently defined after a 3-minute exposure to test aerosols. This brief exposure period is suitable for “mechanical” filters, which become more effective each time they are used, but proves inappropriate for electrostatic filters made of synthetic unwoven fibres, as their performance may diminish over time.
- In real working conditions, the performance of certain electrostatic filters, even those classed “high-efficiency”, can rapidly deteriorate (from the very first time they are used) as the electric charge is gradually neutralised by the trapped dust particles.

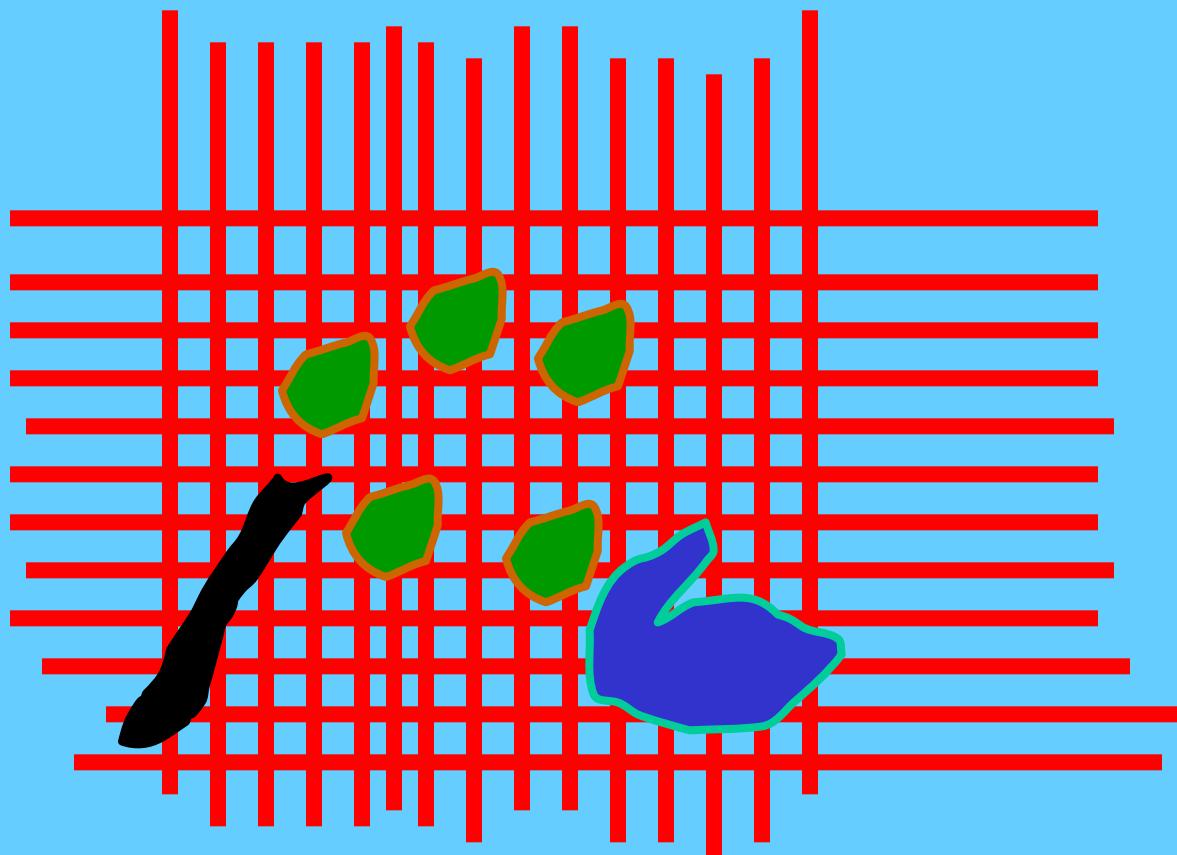
•EUROPEAN SEMINAR ON PERSONAL PROTECTIVE EQUIPMENT

•Saariselkä 19-21 January 2005 Electrostatic filters for respiratory protective devices : an action in progress

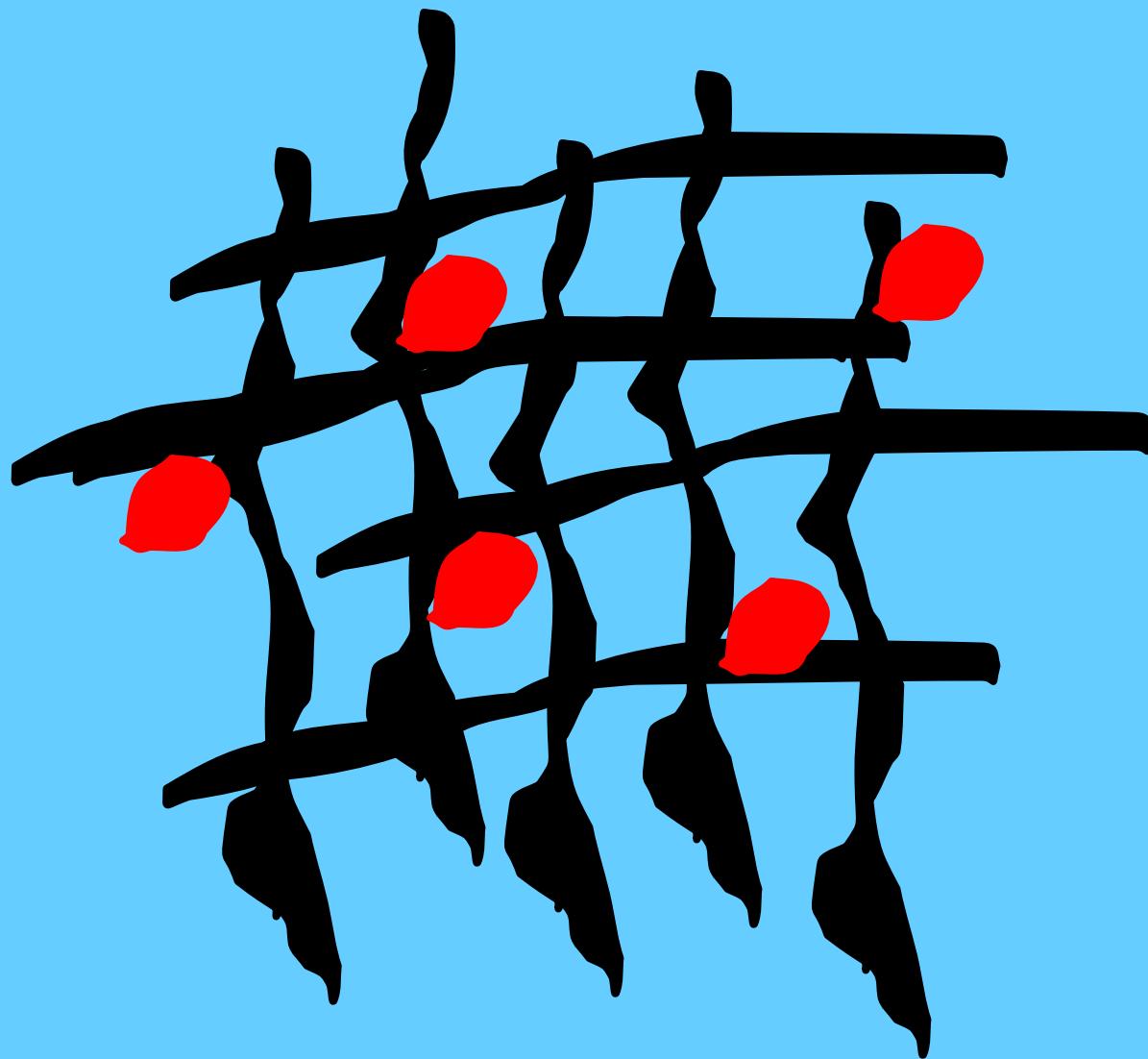
•Pascal Etienne, Patricia Le Frious,French Ministry for Labour, employment and social cohesion

•<http://www.ttl.fi/NR/rdonlyres/99175DFD-757A-4AEB-81EB-021934D9DBB0/0/Etienne.pdf>

Mechanical filter

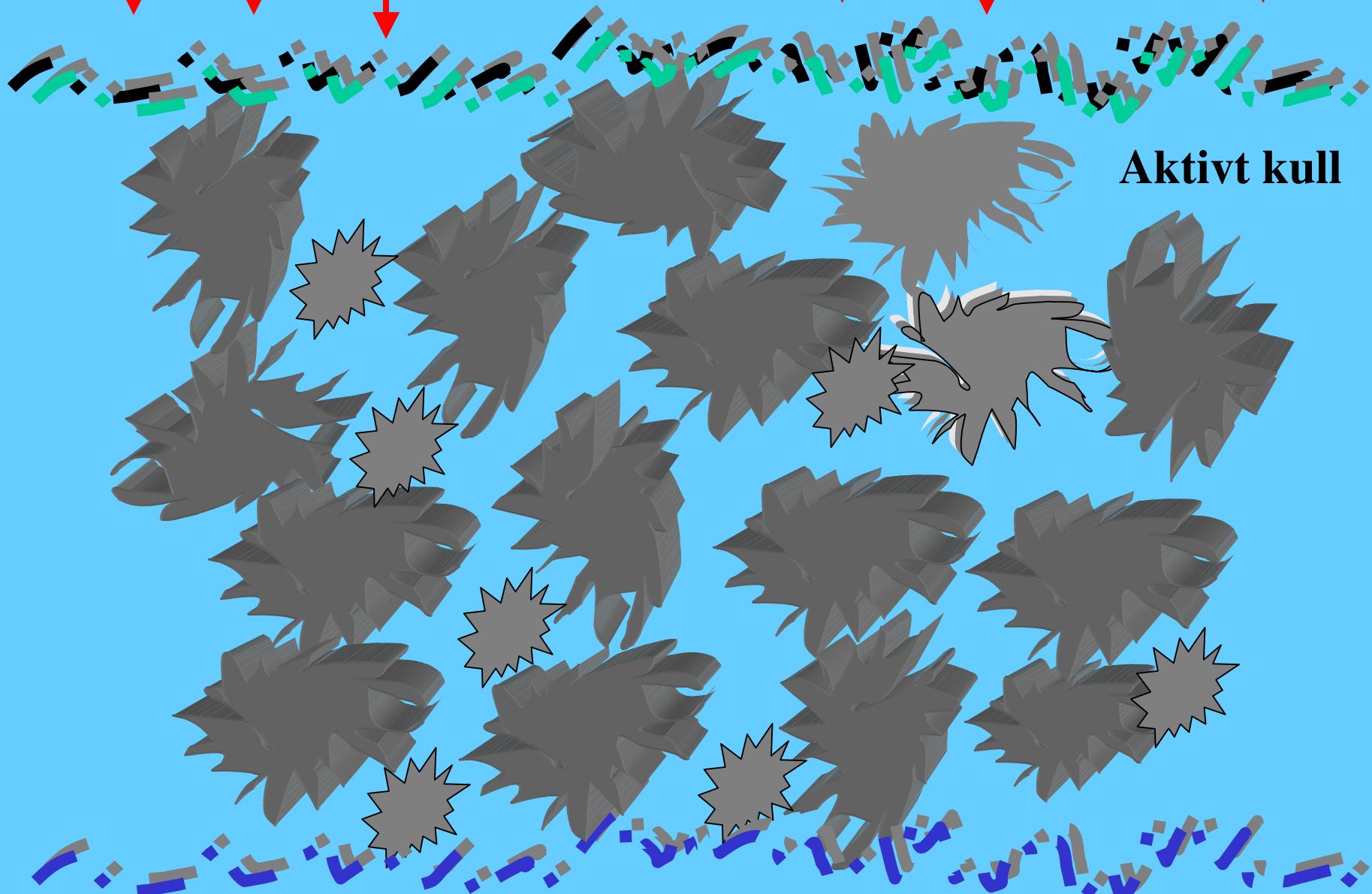


Electrostatic filter



Gassfiltre slipper igjennom aerosoler

Aktivt kull



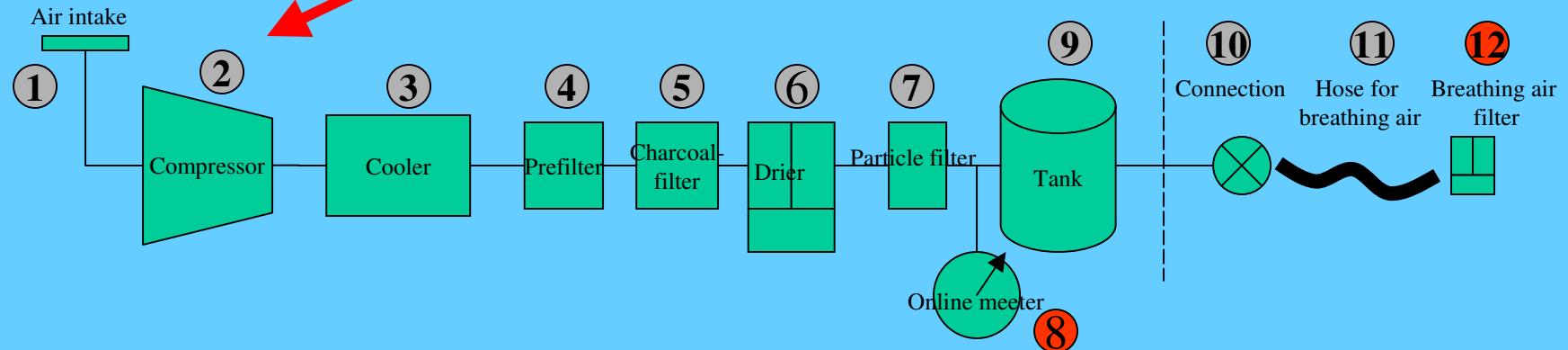


Supplied air



Agricola,
De Re Metallica
1556

The air compressor may be a source for very hazardous exposure from oil leakage and thermal degraded oil



- Review of the breathing system

- <http://www.samarbeidforsikkerhet.no/index.html?infoPage=oppslag.html&siteID=4&id=41&siteID=4&frameID=3&fromTopMenuId=199&&languageCode=EN>

Kilde; Samarbeid for sikkerhet. www.samarbeidforsikkerhet.no

Anbefaling 09/2003 – Gjennomgang av pusteluftsystemene

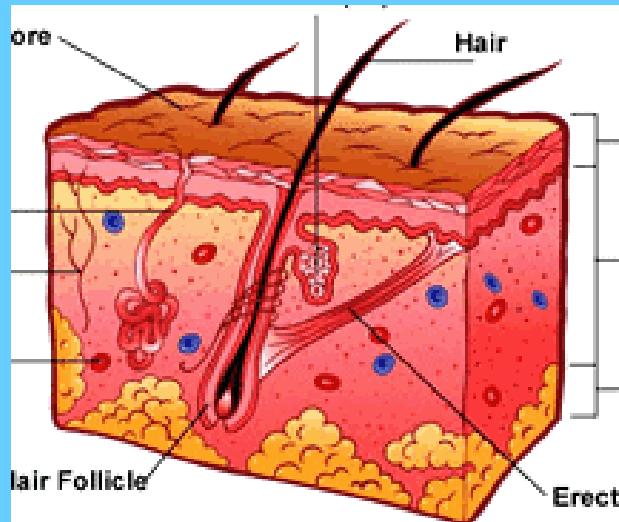
<http://www.samarbeidforsikkerhet.no/index.html?infoPage=oppslag.html&id=41&siteID=&frameID=3&languageCode=NO>





"Hey! Look what Zog do!"

Skin uptake



- For instance, about a teaspoonful of styrene (3 milliliters) splashed on a worker's skin can deliver the same dose as the eight-hour inhalation exposure limit-50 parts per million.
- Just touching a surface contaminated with 1.5 micrograms per square centimeter of the suspected carcinogen acrylamide (a quantity about one-millionth the weight of a paper clip) could equal inhaling one day's permissible exposure limit, 0.03 milligrams per cubic meter.

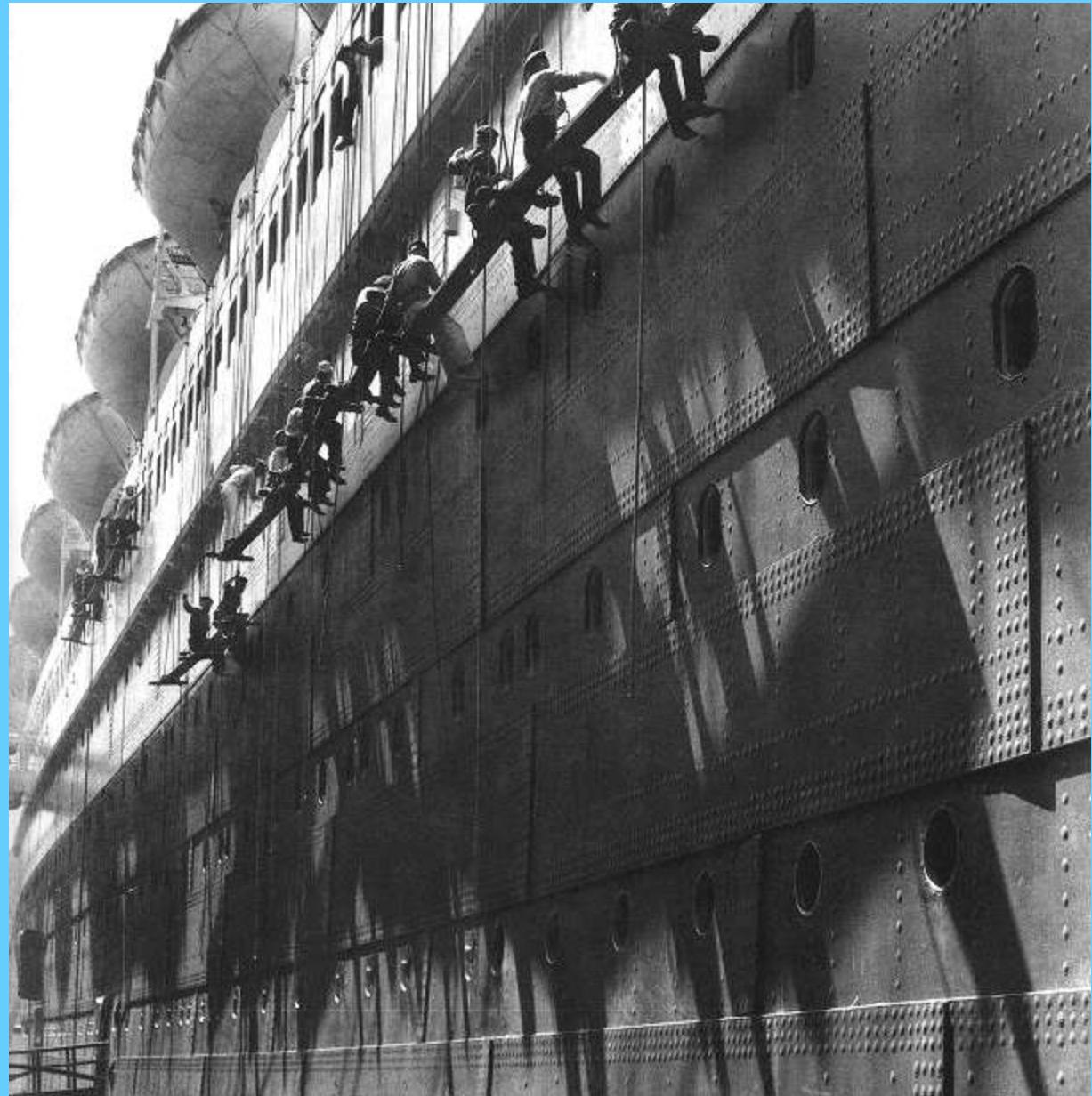
Skin protection, some advices

- **Do:**
- Keep skin clean and healthy. Abraded, irritated, or even sun burnt skin is more susceptible to dermal absorption. Use products tailored to the industrial market, not just cosmetic moisturizers, says Eleanor Fendler, Ph.D., product development manager for skin care company Gojo Industries.
- Keep hands and skin dry. Moisture on the skin, like sweat, can enable permeation, as can high temperatures.
- Choose gloves and protective materials carefully. Some chemicals, like lacquer thinner, can permeate just about any glove when microscopic molecules break through individual molecules of the protective film on the glove. But beware, too, of over protecting. A glove that is too thick or bulky only contributes to exposure risks when workers remove it to perform jobs requiring dexterity or tactility.
- Watch out for glove degradation. Don't rely on a glove once its physical property changes, says Nelson Schlatter of glove maker Ansell Edmont. "Degradation is easy to spot," he says. "The glove will either swell up and get soft, or shrink and harden." Flexing a glove can increase the permeation rate and breakthrough time by ten, according to NIOSH's Boeniger. And, of course, chemicals can penetrate visible holes in gloves.

Skin protection, some advices;

- **Don't:**
- Don't use solvents to clean chemicals off hands. Solvents can damage the skin, making it more readily permeable, says Fendler.
- **Don't put gloves on contaminated hands. Gloves can force penetration of chemicals already on the hands and increase the likelihood of dermal penetration up to five times, according to Boeniger.**
- Don't apply moisturizer or barrier cream to contaminated skin. "If an auto mechanic puts barrier cream on his hands after he changes the oil, he can be causing himself really serious damage by forcing penetration," Fendler says.
- Don't use barrier creams in the place of gloves. Barrier creams can be an addition to a skin protection regime, but studies recommend against substituting them for gloves.

Painting in earlier times



Very hazardous way of painting



Status. April 2002

Find five faults



Sikker strategi

[REDAKTERT] kjenner seg trygg når han maler rør til Statoils Kvitebjørn-dekk. Han følger selskapenes sikkerhetsrutiner. Statoil har laget en kjemikaliestrategi blant annet for å redusere vedlikeholdskostnader til overflatebehandling. Dessuten videreutvikles god praksis for helse, miljø og sikkerhet.

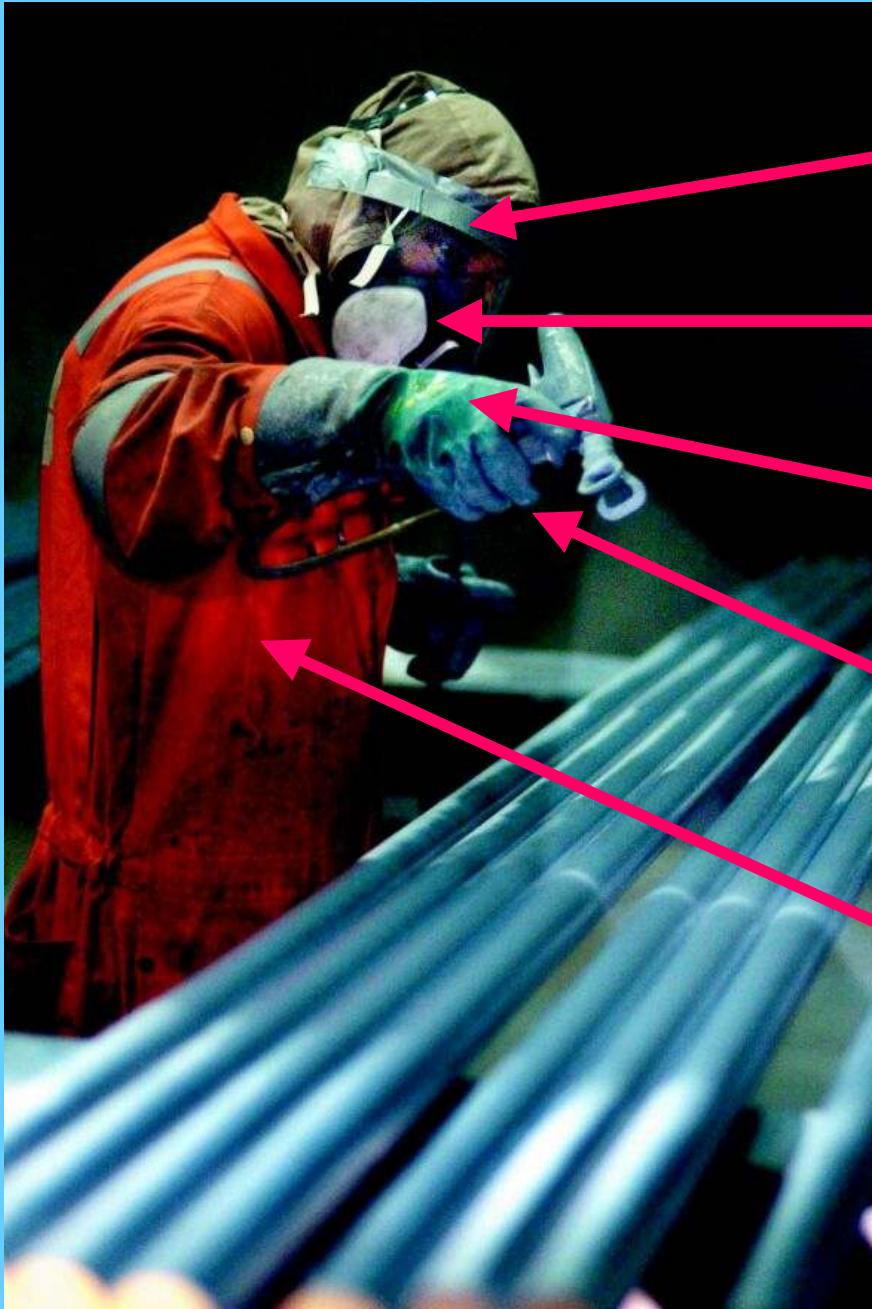
Side 8-9

Jakter på leteområder

Side 2-3

Investeringer bærer frukter

Side 7



Eye protection

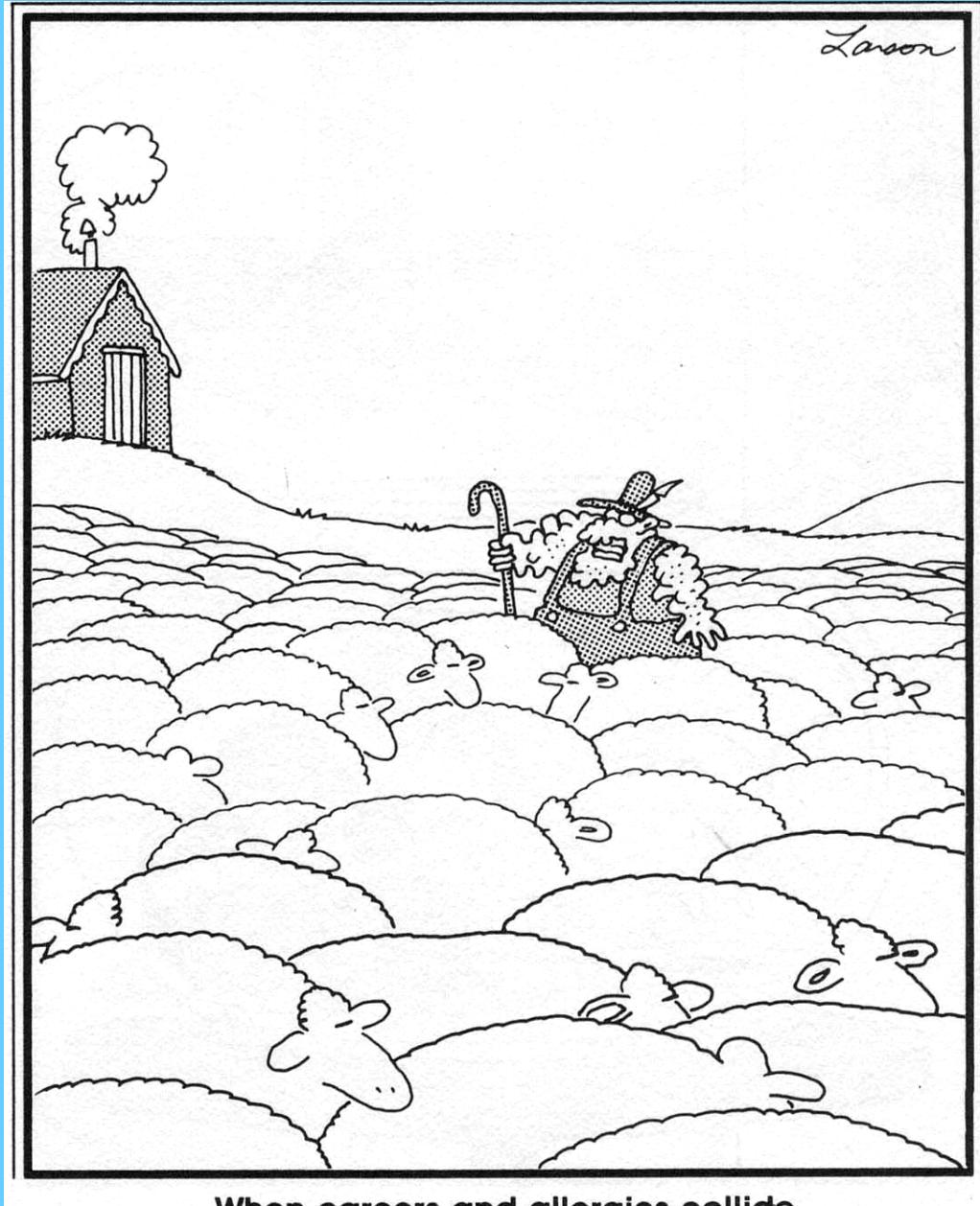
Filter mask use when
spaypainting

Not the right gloves

Safety guards on the
spaying gun missing

Not using a chemical resistant
suit

When
careers and
allergies
collide

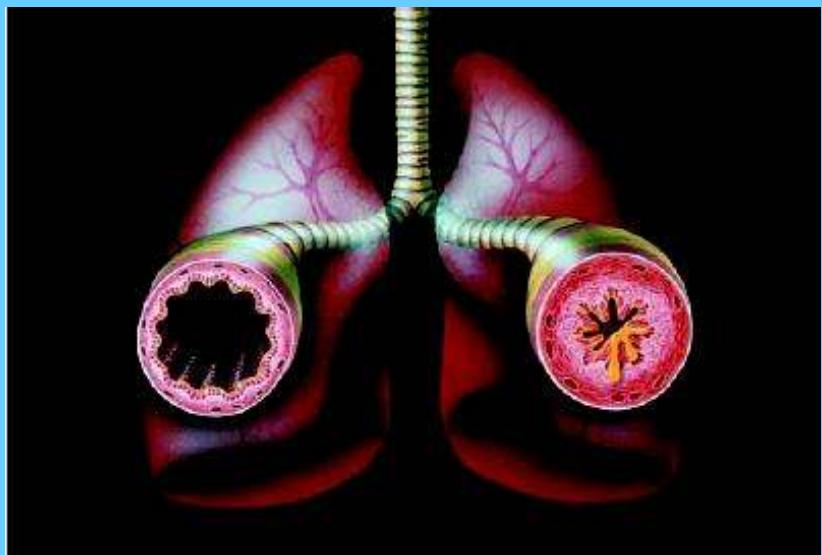


When careers and allergies collide

Epoxy allergy or asthma



- Two conditions that can be developed in a very short time





Får 1,7 millioner i erstatning etter røykforgiftning

– En stor lettelse, sier tidligere Krimpos-betjent Liv Hilde Drugli. Staten må betale henne 1,7 millioner i erstatning etter at hun ble svørt forgiftet på jobben.

KARIN BOHM-PEDERSEN

Gode byrett har også pålagt staten å dekke deler av Druglis ankomkningsmøter med 117 000 kroner. Byrettsdommer Lev Robberstad, Karinsever Hilde, arbeidsgiveren, Krimpos, og staten fort sier at det er gjort til litt for ditt skjønhet i om det er sammenheng mellom brannten for åtte år siden og de alvorlige helseplagene Liv Hilde Drugli fortatt opplyser.

VANT FREM: Liv Hilde Drugli
Foto: KRIMPOS

- Brannrøyk og branngass
- Kortvarig høy eksponering for kjemikalier
 - Maling, sveising, kjemikalieuhelluhell, søl, hudkontakt, ulykker..... unormale driftsforhold, feil verneutstyr

<http://tux1.aftenposten.no/nyheter/riks/d36682.htm>

Immediately Dangerous to Life and Health (IDLH)

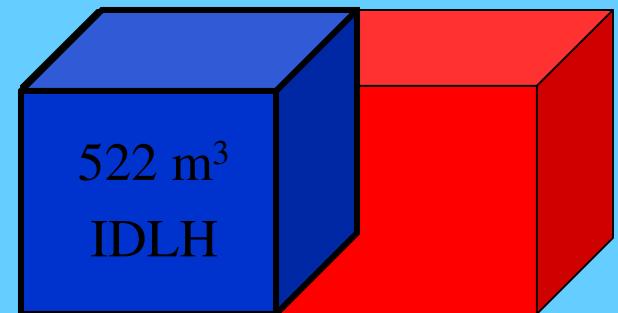
Short term high exposure.

- National Institute of Occupational Safety and Health (NIOSH) has developed a parameter (IDLH) for use to do risk assessment of high level of chemical exposure.
 - *NIOSH Definition*
 - Immediately Dangerous to Life or Health (IDLH): Acute respiratory exposure that poses an immediate threat of loss of life, immediate or delayed irreversible adverse effects on health, or acute eye exposure that would prevent escape from a hazardous atmosphere.
 - *OSHA Definition*
 - Immediately Dangerous to Life or Health (IDLH): An atmosphere that poses an immediate threat to life, would cause irreversible adverse health effects, or would impair an individual's ability to escape from a dangerous atmosphere.
- <http://www.cdc.gov/niosh/idlh/intridl4.html>

How much air to dilute 1 kilogram of toluene to IDLH level?

- The IDLH for toluene is 500 ppm
- $500 \text{ ppm} \times 3,83 = 1915 \text{ milligram/m}^3$
- $1000000 \text{ mg} / 1915 \text{ mg/m}^3 = 522 \text{ m}^3$

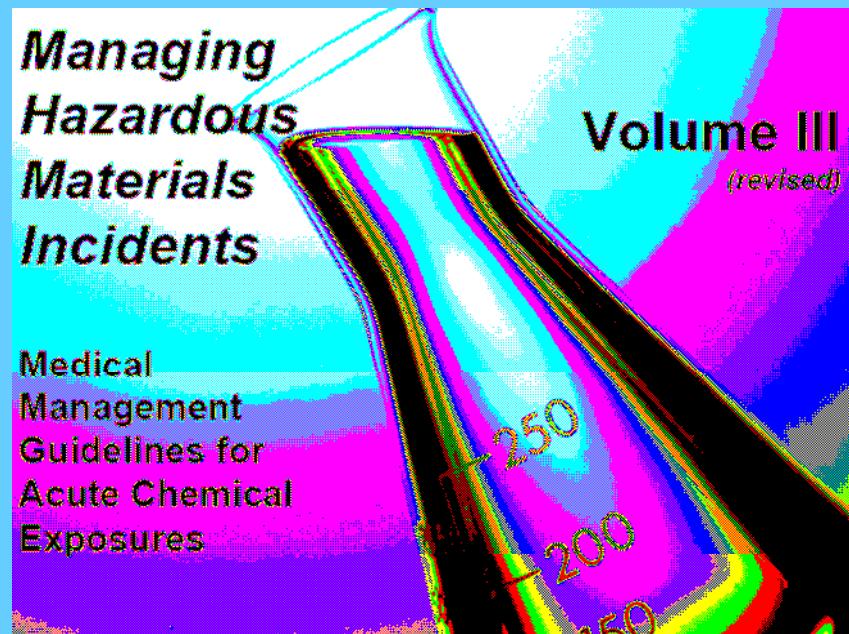
About half the volume of a room of 1000 m³



Comparision between IDLH, OEL, and UEL og LEL

Immediately Dangerous to Life And Health (IDLH) ppm	Occupational Exposure Level	Explosion Level UEL/LEL Vol%
<p>Methanol</p> <p>Xylene</p> <p>Benzene TOLUENE</p> <p>Hydrogen sulfide H₂S (100)</p> <p>Hydrocyanic acid HCN (50)</p> <p>Hydrochloric acid HCl (50)</p> <p>Nitrogen dioxide (NO₂) (20)</p> <p>Formaldehyde (20)</p> <p>Ozone (5)</p> <p>Methyl isocyanat (3)</p> <p>Toluene diisocyanate (2,5)</p> <p>Phosgen (2)</p>	<p>Methanol (100 ppm) S</p> <p>Toluene 25 ppm S</p> <p>Xylene (25 ppm) H</p> <p>Carbon monoxide (25 ppm)</p> <p>Ammonia NH₃ (25 ppm)</p> <p>H₂S (10 ppm) T</p> <p>HCl (5 ppm) T</p> <p>HCN (5 ppm) HT</p> <p>NO₂ (2 ppm) T</p> <p>Benzene (1 ppm) K</p> <p>Hydrofluoric acid (0,8 ppm)</p> <p>Ozone (0,1 ppm)</p> <p>Phosgene (0,05 ppm) T</p> <p>Isocyanates (0,005 ppm) A</p>	<p>Carbon monoxide (74 UEL)</p> <p>Methanol (36 UEL)</p> <p>Methane (15,0 UEL)</p> <p>Carbon monoxide (12,5 LEL)</p> <p>Propane (9,5 UEL)</p> <p>Benzene (7,9 UEL)</p> <p>Xylene (7,0 UEL)</p> <p>Methanol (6,0 LEL)</p> <p>Methane (5,0 LEL)</p> <p>Propane (2,1 LEL)</p> <p>Benzene (1,3 LEL)</p> <p>Toluene 1,1 LEL</p> <p>Xylene (1,0 LEL)</p>

Short term high exposure



- <http://www.atsdr.cdc.gov/MHMI/mhmi-v3-p.pdf#search=%22erpg%20tocp%22>

Emergency Response Planning Guideline (ERPG)

- The Emergency Response Planning Guideline (ERPG) values are intended to provide estimates of concentration ranges where one reasonably might anticipate observing adverse effects as described in the definitions for ERPG-1, ERPG-2, and ERPG-3 as a consequence of exposure to the specific substance.

<http://www.orau.gov/emi/scapa/erpgdefinitions.htm>

Emergency Response Planning Guideline (ERPG) 1

- The ERPG-1 is the maximum airborne concentration below which it is believed that nearly all individuals could be exposed for up to 1 hr without experiencing other than mild transient adverse health effects or perceiving a clearly defined, objectionable odor.

Emergency Response Planning Guideline (ERPG) 2

The ERPG-2 is the maximum airborne concentration below which it is believed that nearly all individuals could be exposed for up to 1 hr without experiencing or developing irreversible or other serious health effects or symptoms which could impair an individual's ability to take protective action.

Emergency Response Planning Guideline (ERPG) 3

- The ERPG-3 is the maximum airborne concentration below which it is believed that nearly all individuals could be exposed for up to 1 hr without experiencing or developing life-threatening health effects.

<u>Chemical (CAS Number)</u>	<u>ERPG-1</u>	<u>ERPG-2</u>	<u>ERPG-3</u>
Dimethyl Disulfide (624-92-0)	0.01 ppm	50 ppm	250 ppm
Dimethylformamide (68-12-2)	2 ppm	100 ppm	200 ppm
Dimethyl Sulfide (75-18-3)	0.5 ppm	1000 ppm	5000 ppm
Epichlorohydrin (106-89-8)	2 ppm	20 ppm	100 ppm
Ethyl Acrylate (140-88-5)	0.01 ppm	30 ppm	300 ppm
Ethylene Oxide* (75-21-8)	N/A†	50 ppm	500 ppm
Ethyldene Norbornene (16219-75-3)	0.2 ppm	100 ppm	500 ppm
Fluorine (7782-41-4)	0.5 ppm	5 ppm	20 ppm
Formaldehyde (50-00-0)	1 ppm	10 ppm	25 ppm
Furfural (98-01-1)	2 ppm	10 ppm	100 ppm
Gluteraldehyde (111-30-8)	0.2 ppm	1 ppm	5 ppm
Hexachlorobutadiene (87-68-3)	1 ppm	3 ppm	10 ppm
Hexafluoroacetone* (684-16-2)	N/A†	1 ppm	50 ppm
Hexafluoropropylene (116-15-4)	10 ppm	50 ppm	500 ppm
Hydrazine (302-01-2)	0.5 ppm	5 ppm	30 ppm
Hydrogen Chloride (7647-01-0)	3 ppm	20 ppm	150 ppm
Hydrogen Cyanide (74-90-8)	N/A†	10 ppm	25 ppm
Hydrogen Fluoride (7664-39-3)**	2 ppm	20 ppm	50 ppm
Hydrogen Peroxide (7722-84-1)	10 ppm	50 ppm	100 ppm
Hydrogen Selenide (7783-07-5)	N/A	0.2 ppm	2 ppm
Hydrogen Sulfide (7783-06-4)	0.1 ppm	30 ppm	100 ppm

Blåsyre
Saltsyre
Hydrogensulfid

<u>Chemical (CAS Number)</u>	<u>ERPG-1</u>	<u>ERPG-2</u>	<u>ERPG-3</u>
Phenol (108-95-2)	10 ppm	50 ppm	200 ppm
Phosgene (75-44-5)	N/A	0.2 ppm	1 ppm
Phosphine (7803-51-2)	NA†	0.5 ppm	5 ppm
Phosphorus Pentoxide (1314-56-3)	1 mg/m ³	10 mg/m ³	50 mg/m ³
Phosphorus Trichloride (7719-12-2)	0.5 ppm	3 ppm	15 ppm
Propylene Oxide (75-56-9)	50 ppm	250 ppm	750 ppm
Sodium Hydroxide (1310-73-2)	0.5 mg/m ³	5 mg/m ³	50 mg/m ³
Stibine (7803-52-3)	ID‡	0.5 ppm	1.5 ppm
Styrene (100-42-5)	50 ppm	250 ppm	1000 ppm
Sulfur Dioxide (7446-09-5)	0.3 ppm	3 ppm	15 ppm
Sulfuric Acid (Oleum [8014-95-7], Sulfur Trioxide [7446-11-9], and Sulfuric Acid [7664-93-9])	2 mg/m ³	10 mg/m ³	30 mg/m ³
Tetrachlorosilane (10026-04-7)	0.75 ppm	5 ppm	37 ppm
Tetraethoxysilane (78-10-4)	25 ppm	100 ppm	300 ppm
Tetrafluoroethylene (116-14-3)	200 ppm	1000 ppm	10,000 ppm
Tetrahydrofuran (109-99-9)	100 ppm	500 ppm	5000 ppm
Tetramethoxysilane (681-84-5)	NA†	10 ppm	20 ppm
Thionyl Chloride (7719-09-7)	0.2 ppm	2 ppm	10 ppm
Titanium Tetrachloride (7550-45-0)	5 mg/m ³	20 mg/m ³	100 mg/m ³
Toluene (108-88-3)	50 ppm	300 ppm	1000 ppm
Toluene 2,4- (2,6-) Diisocyanate (TDI) (584-84-9)	0.01 ppm	0.15 ppm	0.6 ppm
1,1,1-Trichloroethane (71-55-6)	350 ppm	700 ppm	3500 ppm

Fosgen

Svoveldioksid

Toluendiisocyanat
0,01 , 0,15 0,6

Hot Work



Rivelin edge in the Peak district is an ideal training ground for extreme ironing

Bilde; Extreme Ironing. www.extremeironing.com

Hot work. Welding, grinding, flame cutting creates many very toxic compounds



Do you work with ISOCYANATES and POLYURETHANE?

http://www.prevent.se/doc_pdf/verktyg/pdf/isoeng_broschyr.pdf

http://www.prevent.se/doc_pdf/verktyg/pdf/isoeng_diagnos.pdf

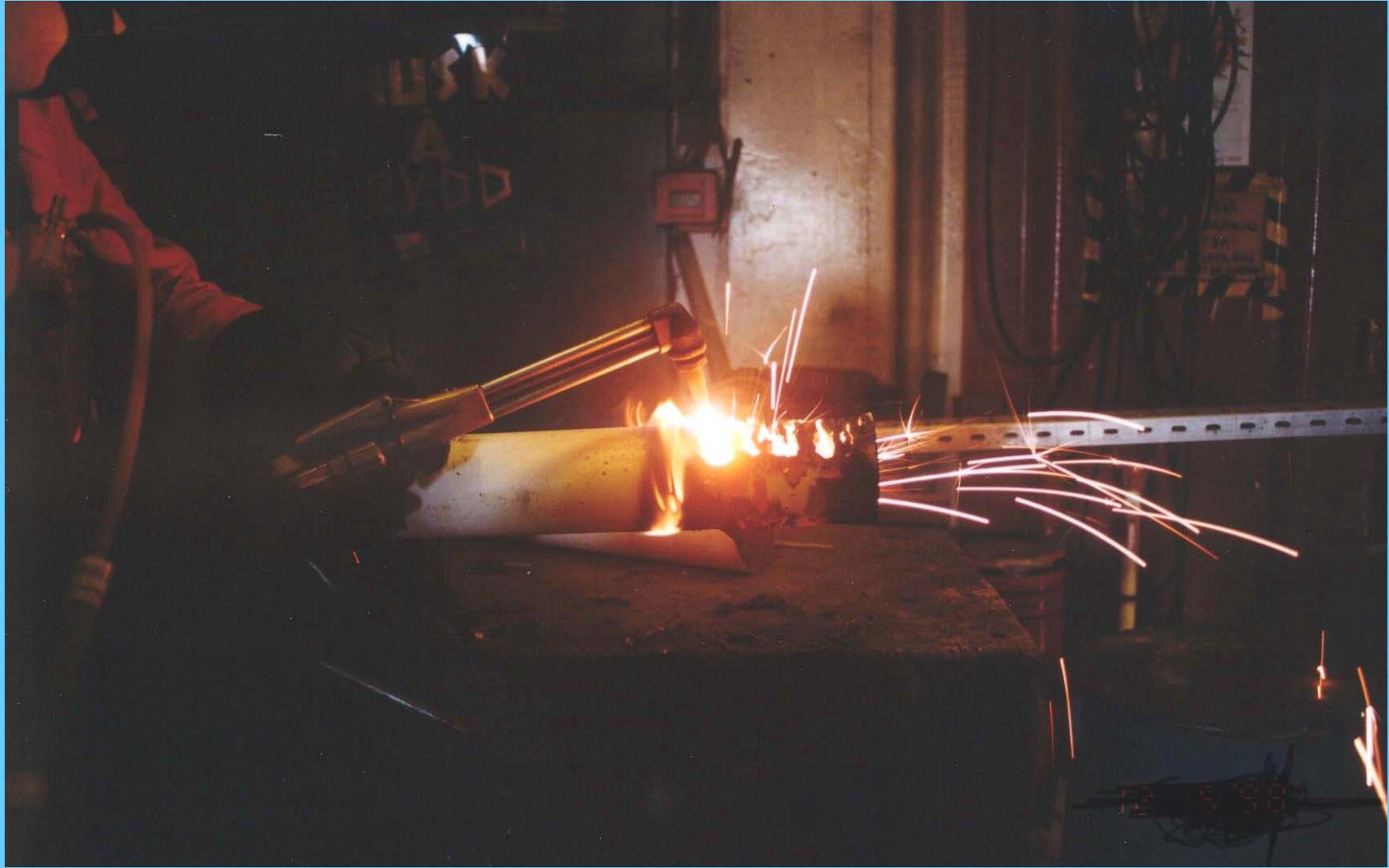
NIOSH Safety and Health Topic: Isocyanates

<http://www.cdc.gov/niosh/topics/isocyanates/>

Foto: Halvor Erikstein

Before





- Cutting and heating of painted surfaces, what can happen?



Thermal degradation of polyurethane – general mechanism

Polyurethane → isocyanate → amine → nitrils → nitrous oxides (NOx)



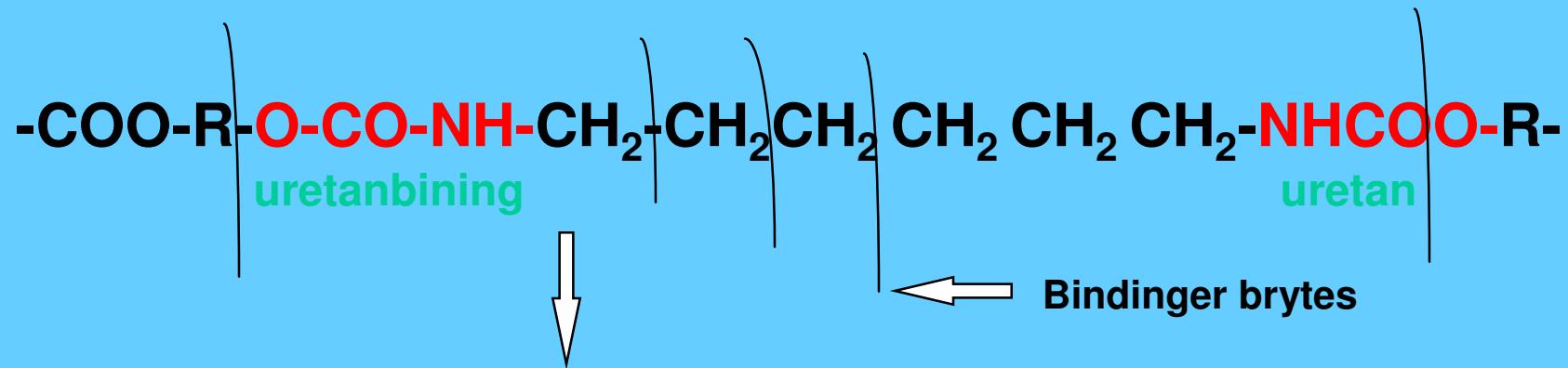
Example HDI based

HDI-dekomp. → Methyl isocyanat → Methyl amine → Acetonitril → Hydrogen cyanidr
 CH_3NCO $\text{CH}_3\text{-NH}_2$ $\text{CH}_3\text{-CN} /$ HCN

Fremstilling og bruk av polyuretanprodukter (Isocyanater)
<http://www.arbeidstilsynet.no/publikasjoner/brosjyrer/bros536.html>

Les mer om isocyanater på:
<http://www.arbeidstilsynet.no/sok.html?q=isocyanater>

Thermal degradation of HDI-based polyurethane



$\text{CH}_3\text{-NCO}$

$\text{CH}_3\text{Ch}_2\text{-NCO}$

$\text{CH}_3\text{CH}_2\text{CH}_2\text{-NCO}$

$\text{OCN-(CH}_2)_6\text{-NCO}$

+

Combinations of isocyanates and amines

And other groups. Very complex chemistry

Methyl isocyanate

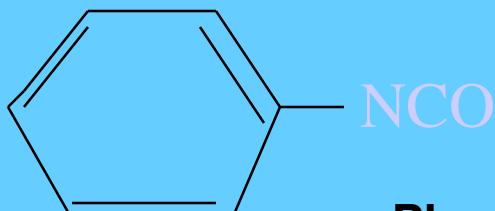
Ethyl isocyanat

Propy lisocyanat

Hexamethylene diisocyanate (HDI)

Compounds identified in the smoke from heated polyurethane top coat

	OEL	mikrogram/m3 Measured
CH3-NCO	12	470
CH3CH2-NCO	15	400
CH3CH2CH2-NCO	20	140
OCN-CH2CH2CH2CH2CH2-NCO	30	340



Phenyl isocyanate

3

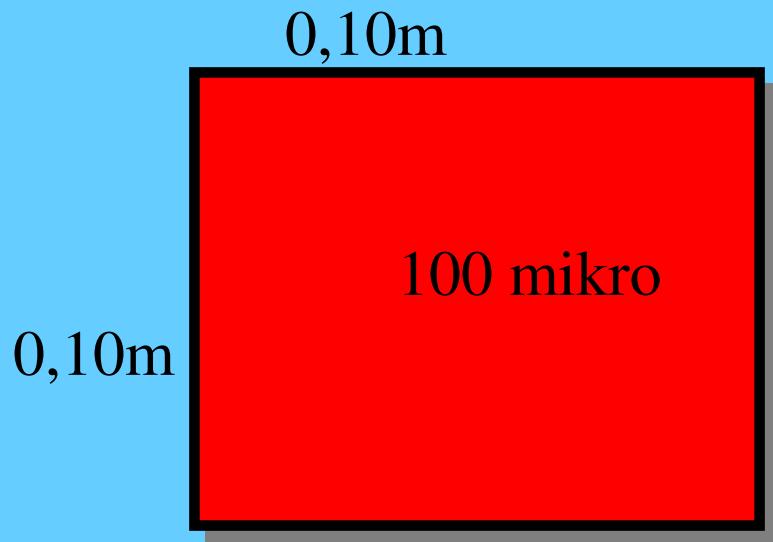
Occupational exposure level (OEL) and times exceeding the OEL

	OEL	Measured	Times
MIC	12	470	39
EIC	15	400	27
PIC	20	140	7
HDI	30	340	11
		Total	84

OEL 0,005 ppm

– UK OEL (0,02 mg(-NCO)/m³)

Example on amounts of isocyanate from heating polyurethane based paint



Thickness 100 mikrometer

Density = 1 mg/m³

$$0,10m \times 0,10m \times 0,000001m \times 1\text{mg}/\text{m}^3 \\ = 1000\text{ mg}$$

OEL MDI 0,05mg/m³.

1% of the paint is degraded back to isocyanate

A degrading rate of 1% from 1000 mg gives $(1000\text{mg} \times 1\%) = 10\text{mg}$.

Air dilution needed for the $10\text{mg}/0,05\text{mg}/\text{m}^3 =$

200 m³



Corporate Man stands up for something that he knows to be right.

What can you be met with, if your are getting sick
from the working environment?
A coil shoulder



Read DAGBLADSERIEN "Oljemarerittet"
<http://www.dagbladet.no/nyheter/87698.html>





*"What this lab really needs
is better ventilation"*

©1998 BoGrace

Kjemisk eksponering skal reduseres ved hjelp av tekniske tiltak



Foto; Olesen Consult www.olesen-hvac.no